

DIPLOMA IN LEATHER TECHNOLOGY

Sl. No.	Course Code	Course Name	Teaching Scheme					Examination Scheme							
			Instruction periods per week			Total Periods per semester	Credits	Continuous internal evaluation			Semester end examination				
			L	T	P			Mid Sem1	Mid Sem 2	Internal Evaluation	Max marks Min marks	Total Marks	Min marks for passing including internal		
1	18LT-201F	Communicative English	2.4	1.6	0	60	3	20	20	20	40	14	100	45	
2	18LT-202F	Engineering mathematics	2.4	1.6	0	60	3	20	20	20	40	14	100	45	
3	18LT-203F	Applies Physics	2.4	1.6	0	60	3	20	20	20	40	14	100	45	
4	18LT-204F	Engineering Chemistry & Environmental Studies	2.4	1.6	0	60	3	20	20	20	40	14	100	45	
5	18LT-205C	Introductory Post Tannages	2.4	1.6	0	60	3	20	20	20	40	14	100	45	
6	18LT-206P	Adv Engineering Drawing Practice	1	0	2	45	1.5	20	20	20	40	20	100	45	
7	18LT-207P	Adv computer aided Drafting	1	0	2	45	1.5	20	20	20	40	20	100	45	
8	18LT-208P	Post Tanning of Leathers Practice	1	0	2	45	1.5	20	20	20	40	20	100	45	
9	18LT-209P	Applied Science Lab Practice	1	0	2	45	1.5	20	20	20	40	20	100	45	
10	18LT-210P	Information Technology Lab Practice	1	0	2	45	1.5	20	20	20	40	20	100	45	
11		Academic activities	0	0	7	105	2.5	0	0	Rubrics		--	--	-	
		TOTAL	17	8	17	630	25	200	200	200	400	150	1000	450	

Department of Technical Education

State Board of Technical Education & Training, Telangana

Course Title : Advanced English	Course Code : 18 LT-201F
Semester : II	Course Group : Foundation
Teaching Scheme in Periods (L:T:P): 36:24:0	Credits : 3
Methodology : Communicative Language Teaching + Assignments	Total Contact Hours : 60 periods
CIE : 60 Marks	SEE : 40 Marks

Prerequisites: Basic knowledge of English Language

COURSE OUTCOMES

	At the end of the course the students will have the ability to:
201.1	learn homonyms and one word substitutes and use them in professional interaction
201.2	listen for specific purpose and use appropriate prepositions.
201.3	acquire values through stories and reports
201.4	write resumes, reports and make notes
201.5	work in pairs and groups confidently
201.6	analyse errors and make communication flawless

CO-PO Matrix

CO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	Mapping POs
201.1	2	2	2	--	1	--	--	3	3	3	1,2,3,5,8,9,10
201.2	2	2	1	2	--	--	--	3	3	3	1,2,3,4,8,9,10
201.3	2	2	--	1	1	1	2	3	3	3	1,2,4,5,6,7,8,9,10
201.4	2	2	2	2	1	--	2	3	3	3	1,2,3,4,5,7,8,9,10

201.5	2	2	2	--	1	--	2	3	3	3	1,2,3,5,7,8,9,10
201.6	2	2	--	--	--	--	--	3	3	3	1,2,8,9,10

COURSE CONTENTS

UNIT – 1 SPEAKING

Duration: 10 Periods

1. Expressing Obligation
2. Fixing and Cancelling Appointments
3. Extending and Accepting Invitations
4. Giving Instructions
5. Asking for and Giving Directions

UNIT - 2: LISTENING

Duration: 6 Periods

6. The Here and Now!

UNIT –3: VOCABULARY

Duration: 6 Periods

7. How to Learn a New Word
8. Synonyms, Antonyms and One word Substitutes

UNIT -4: GRAMMAR

Duration: 12 Periods

9. Reported Speech
10. Error Analysis - I
11. Error Analysis - II
12. Error Analysis - III

UNIT - 5: READING

Duration: 6 Periods

13. An Environmental challenge
14. The Will to Succeed
15. Waiting for Mr. Clean

UNIT- 6: WRITING

Duration: 20 Periods

16. Data Interpretation- I
17. Data Interpretation- II
18. Data Interpretation- III
19. Writing a Resume
20. Writing a Cover Letter
21. Note Making
22. Writing a Report

Specific Learning Outcomes:

On completion of the course the students will be able to:

- express obligation, fix and cancel appointments, extend –accept and decline invitations.
- give instructions and directions
- identify and use prepositions
- learn homonyms and use one word substitutes
- read and understand main ideas and answer the questions
- understand and write reported speech
- identify and correct common errors
- interpret data
- learn to prepare cover letter and resume
- make notes and write reports

Internal evaluation

Test	Units	Marks	Pattern
Mid Sem 1	Speaking Listening	20	Part A 5 Short answer questions Part B 2 Essay questions out of 3 Questions Part C 2 Essay questions out of 3 Questions
Mid Sem 2	Vocabulary Grammar	20	Part A 5 Short answer questions Part B 2 Essay questions out of 4 Questions Part C 2 Essay questions out of 3 Questions

Slip Test 1	Speaking Listening	5	2 Essay Questions out of 3 Questions
Slip Test 2	Vocabulary Grammar	5	2 Essay Questions out of 3 Questions
Assignment	One assignment per one semester	5	Different group assignments of Higher order Questions that develop problem solving skills and critical thinking should be given
Seminars	One seminar per one semester	5	
	Total	60	

Suggested Student Activities:

- Listen to a song and answer the questions
- Listen to a passage/conversations/dialogues/speeches and answer the questions
- Group Discussions
- Student Presentations
- Seminars
- Talk about a movie/review
- Talk about a book
- Narrating a story
- Chain stories
- JAM on topics like environment, pollution, ethics, morals, responsibilities of citizens
- Speak about incidents/events/memories/dreams/role model
- Interview with famous personalities
- Cricket commentary
- Reading for main ideas
- Reading for specific details
- Summarizing
- Picture description
- Writing a recipe
- Surprise test
- Compare and contrast two people/pictures/news items/ideas etc
- Surveys
- Filling forms
- e-mail etiquette

Textbook: English for Polytechnics

REFERENCES:

1. Practical English Grammar by A.J Thomson and A.V. Martinet
2. A Course in Phonetics and Spoken English by J. Sethi and P.V Dhamija
3. Word Power Made Easy by Norman Lewis
4. Keep Talking by Friederike Klippel
5. More Grammar Games by Mario Rinvoluceri and Paul Davis
6. Essential English Grammar by Raymond Murphy
7. Spoken English-A Self Learning Guide to Conversation Practice by V Sasi Kumar

e-learning:

1. www.duolingo.com
2. www.bbc.co.uk
3. www.babbel.com
4. www.merriam-webster.com
5. www.ello.org
6. www.lang-8.com
7. youtube.com
8. Hello English(app)
9. mooc.org
10. <https://onlinecourses.nptel.ac.in>

DISTRIBUTION OF QUESTIONS/MARKS FOR SEMESTER MID/END EXAMINATION

Module	Unit Name	No. of Periods	NUMBER OF QUESTIONS TO BE CONSIDERED										UNIT WISE WEIGHTAGE	(MS+EE) WEIGHT AGE
			R		U		A		MARKS WEIGHTAGE					
			MID	END	MID	END	MID	END	MS-I	MS-II	MS-III	END EXAM		
PART-A	Speaking	10	3	1	2	1	2	1	36	-	-	17	53	74
	Listening	6	2	1	1	0	1	0	19	-	-	02	21	
PART-B	Vocabulary	6	2	1	1	0	1	0	-	19	-	02	26	74
	Grammar	12	3	1	2	1	2	1	-	36	-	17	48	
PART-C	Reading	6	3	1	0	0	1	0	-	-	16	2	18	72
	Writing	20	2	0	3	1	2	1	-	-	39	15	54	
TOTAL		60	15	05	9	3	9	3	55	55	55	55	220	220
											110			

LEGEND	R: Remembering
	U: Understanding
	A: Applying

Semester End Examination Marks Distribution

	Short answer	Essay	Marks
Part A	10	0	20
Part B			
Group 1	0	2/3	10
Group 2	0	2/3	10
Part C			
Group 1	0	2/3	20
Group 2	0	2/3	20
Total	10	8/12	80

Mid Sem Examination Marks Distribution

	Short answer	Essay	Marks
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Part A	5	0	10
Part B	0	2/3	10
Part C	0	2/3	20
Total	5	4/6	40

Mid Sem Examination marks distribution

	Short answer	Essay	Marks
Part A	5	0	10
Part B	0	2/3	10
Part C	0	2/3	20
Total	5	4/6	40

State Board of Technical Education, Telangana State
Model Paper- 18EC201F (Advanced English)
Mid Sem-I

Time : 1 ½ Hours

Total Marks : 40

PART – A

5 X 2 = 10

Instructions:

- i) *Answer all the following questions:*
- ii) *Each question carries two marks.*

1. Write two sentences, one with 'must' and another with 'have to', to express obligations.
2. Fill the blanks with suitable prepositions.
 - a) He came home _____ a car.
 - b) Kiran mixed coffee _____ a spoon.
3. How do you invite your neighbor to attend a seminar on global warming?
4. Fix an appointment with the dentist at 5.30 p.m.
5. Fill the blanks with the suitable prepositions given below:

Among, between, by, with, from, at, for

- a) She distributed sweets _____ her two brothers.
- b) Mohan died _____ cancer.

PART- B

2 X 5 = 10

Instructions:

- i. *Answer any two questions.*
- ii. *Each question carries five marks.*

6. Write a dialogue between you and the reception about the cancellation of an appointment you have with the doctor.
7. Give directions to your friend to reach to the park from your house.
8. Write a paragraph describing your polytechnic using at least five prepositions.

PART- C

2 X 10 = 20

Instructions:

- i. *Answer any two questions.*
- ii. *Each question carries ten marks.*

9. Fill the blanks with the suitable prepositions.
- He looked ____ me.
 - Listen ____ my instructions carefully.
 - Geetha suffered _____ fever.
 - Bhagya threw a stone _____ the well.
 - Prathap kept a ladder _____ the wall.
 - We played cricket _____ five hours.
 - My books were stolen _____ Kiran.
 - We will go to library _____ 15th of this month.
 - Shailaja has been reading a novel _____ 10.00 a.m.
 - We bought this television _____ 2014.
10. Give instructions on how to send an e-mail to your friend.
11. a). Fix an appointment with your M.L.A. next Sunday at 4.00 p.m. to discuss the problems in your village.
- b). Cancel the same appointment as you are going to leave for Adilabad on some urgent personal work

State Board of Technical Education, Telangana State
Model Paper- 18EC201F (Advanced English)
Mid Sem-II

Time : 1 ½ Hours

Total Marks : 40

PART – A

5 X 2 = 10

Instructions:

- Answer all the following questions:*
- Each question carries two marks.*

- Write one word substitutes for the following expressions.
 - A place where books are available to be borrowed and for reference.
 - That which cannot be heard.
- Write the synonyms of the following words:
 - Rich
 - Happy
- Change the following into indirect speech.
 - Vinod said, "I have gone to Bhadrachalam yesterday."
 - Gopal said to Mamatha, "I will play cricket tomorrow."
- Change the following onto direct speech.
 - David said to Madhavi, "Give me your calculator now."
 - Jayanth said to Fathima, "Where are you going?"

5. Correct the words given in italics in the following sentences.
- Lalitha *go* to Nanded tomorrow.
 - Adarsh sat *besides* Vikas.

PART-B

2 X 5 =10 marks

Instructions:

- Answer any two questions.
- Each question carries 5 marks.

6. How should a new word be learnt?
7. Correct the following sentences.
- It is very hot to go outside.
 - Shiva works hardly.
 - She is more cleverer than Bhaskar.
 - Vidya is senior than Indira.
 - Praveen is angry on his sister.
8. Change the following into indirect speech.
- She said to him, “When will you go to temple?”
 - Radhika said to her brother, “Will you pay the examination fee tomorrow?”
 - Vasu said, “I do not eat non-vegetarian food.”
 - Rafi said to Mathews, “Show me your record.”
 - Pramod said to Sandhya, “My brother will bring fruits tomorrow from Karimnagar.”

PART-C

2 X 10 =20 marks

Instructions:

- Answer any two questions.
- Each question carries ten marks.

9. Mention any five ways of learning a new word.
10. Change the following dialogue into a paragraph.
- Ajay: Hai Sudha! How are you?
Sudha: I am fine. How are you?
Ajay: I am fine too. Where are you going now?
Sudha: I am going to market.
Ajay: What do you want to buy there?
Sudha: I want to buy fruits and vegetables.
Ajay: Do you know the mobile number of Suresh?
Sudha: No. I don't have his mobile number. Why do you need his number?
Ajay: I want to invite him for my sister's birthday.

11. Correct the following sentences.
- a. Every student has to bring their textbook tomorrow.
 - b. She is having a house.
 - c. We are living in this house for the last ten years.
 - d. This machine works perfectly.
 - e. He is the taller student in my class.
 - f. Music classes begin from 27th July.
 - g. She is weak and she can run fast.
 - h. The door was painted by a small brush.
 - i. Where your brother is studying?
 - j. He awaited for the bus here yesterday.

State Board of Technical Education, Telangana State
C18-Semester End Examination (SEE)
Model Paper- 18Common201F (Advanced English)

Time: 3 Hours

Total Marks: 80

PART – A

Instructions:

10 X 2 = 20

- i. Answer all the following questions.*
 - ii. Each question carries two marks.*
1. Fill the blanks with the suitable expressions of obligation:
 - a) We _____wear helmet while riding a two wheeler.
 - b) A student _____ be in time to college.
 2. Fill the blanks with suitable prepositions:
 - a) He went to polytechnic ____ a bicycle.
 - b) They have been waiting _____ a bus since 8.00 a.m.
 3. Write the antonyms of the following words:
 - a) Legal

b) Honest

4. Change the following sentences into direct speech:
 - a. He said that he had a beautiful house.
 - b. The visitors thanked the guide.
5. What do you write about the following ones in your resume?
 - a. Your skills
 - b. Your work experience
6. What do you write about the following ones in your resume?
 - a. Your interests and activities
 - b. Your educational qualifications
7. Read the following paragraph and answer the questions given in questions no. 7 and 8.

Subhas Chandra Bose was born in a Bengali Kayasth family on January 23, 1897 in Cuttack (Odiya Baazar), Orissa, to Janakinath Bose, and Prabhavati Devi. He was the ninth child of 14. He studied in an Anglo school at Cuttack (now known as Stewart School) until standard 6. He then shifted to Ravenshaw Collegiate School of Cuttack. From there he went to the prestigious Presidency College where he studied briefly. His nationalistic temperament came to light when he was expelled for assaulting Professor Oaten for his anti-India comments.

His high score in the Civil Service examinations meant an almost automatic appointment. He then took his first conscious step as a revolutionary and resigned the appointment on the premise that the best way to end a government is to withdraw from it. At the time, Indian nationalists were shocked and outraged because of the Amritsar massacre and the repressive Rowlatt legislation of 1919. Returning to India, Bose wrote for the newspaper Swaraj and took charge of publicity for the Bengal Provincial Congress Committee. His mentor was Chittaranjan Das, spokesman for aggressive nationalism in Bengal. Bose worked for Das when the latter was elected mayor of Calcutta in 1924. In a roundup of nationalists in 1925, Bose was arrested and sent to prison in Mandalay, where he contracted tuberculosis.

Answer the following questions.

- a. Where was Subhas Chandra Bose born?

- b. Who were his parents?
8. Answer the following questions
- a. Why was Bose expelled from Presidency College?
 - b. Why was he sent to Mandalay?
9. Read the following paragraph and answer the questions given in questions no. 9 and 10.

Dr. Rajendra Prasad, son of Mahadev Sahai, was born in Zeradei village, in the Siwan district of Bihar, on 3 December 1884. He was the youngest in a large family, & was close to his mother and eldest brother. He was known as “Rajen” to his family and friends. His father, Mahadev Sahay, was a scholar of both the Persian and Sanskrit languages, while his mother, Kamleshwari Devi, was a religious woman. Zeradei’s population was diverse, with both Muslims and Hindus living in relative harmony.

When Rajendra Prasad was five years old, his parents put him under a Mawlawi, an accomplished Muslim scholar, to learn the Persian language, followed by Hindi and arithmetic. After the completion of traditional elementary education, Rajendra Prasad was sent to the Chhapra District School. At the age of 12, Rajendra Prasad was married to Rajavanshi Devi. He, along with his elder brother Mahendra Prasad, then went on to study at T.K. Ghosh’s Academy in Patna.

Since childhood, Rajendra Prasad was a brilliant student. He placed first in the entrance examination to the University of Calcutta and was awarded Rs.30 per month as a scholarship. In 1902, Rajendra Prasad joined the Presidency College. He was initially a student of science and his teachers included Jagadish Chandra Bose and Prafulla Chandra Roy. Later he decided to focus on the arts. Prasad lived with his brother in the Eden Hindu Hostel

Answer the following questions:

- a. Where was Rajendra Prasad born?
 - b. What was he known as?
10. Answer the following questions.
- a. Where did he learn the Persian language?

- b. Where was he awarded Rs. 30 per month as a scholarship?

PART- B

4 X 5 = 20

Group 1

2 X 5 = 10

Instructions: 1. Answer any two of the following questions.

2. Each question carries five marks.

11. Write instructions on how to prepare tea.
12. Write a cover letter to the Managing Director, Vijaya Cement Works, Godavarikhani as you wish to apply for the post of Assistant Executive Engineer.
13. Correct the following sentences.
- They congratulated Aravind for his success.
 - Though Anand is poor, but he is honest.
 - I wish I have a laptop.
 - Nafeesa and me are playing shuttle badminton.
 - Hari is having a car.

Group 2

2 X 5 = 10

Instructions: 1. Answer any two of the following questions.

2. Each question carries five marks.

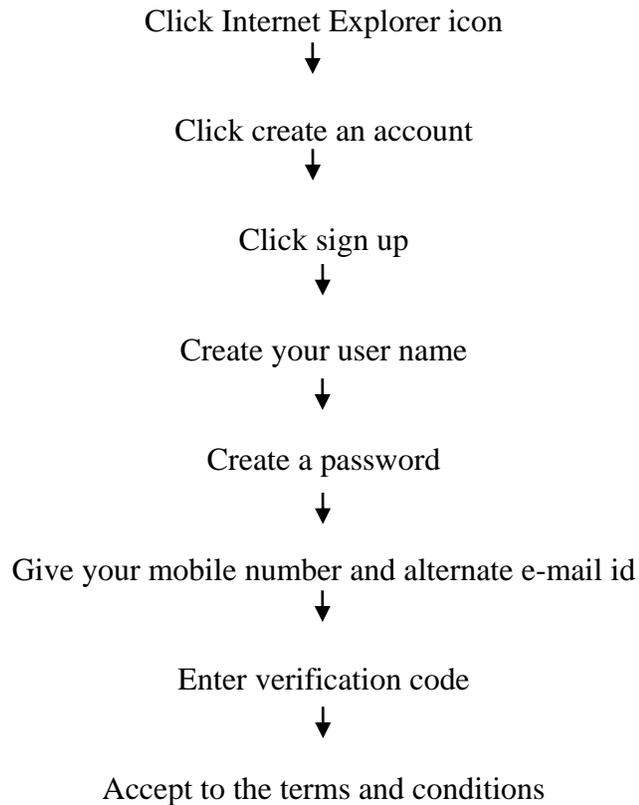
14. Write a report on the industry you have visited last week.
15. Observe the following table and write a paragraph analyzing the information given in it.

Table showing the number of students admitted in different courses in Private Engineering colleges in the past four years.

Year	ECE	EEE	Mechanical	Civil	CSE
2017	54065	36255	21600	34000	13436
2016	49008	36255	20900	29000	22687
2015	45032	36255	20600	14500	32008

2014	38060	36254	20300	14500	38065
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16. Observe the following flow chart and write a paragraph describing the steps involved in creating an e-mail.



PART-C

4 X10 = 40

Group 1

2 X 10 = 20

Instructions: 1. Answer any two of the following questions.

2. Each question carries ten marks.

17. Write a resume to apply for the post of AEE in the Department of Tribal Welfare, Government of Telangana.

18. Write instructions for the following:

- a. Opening an account in a bank.
- b. Taking a bus pass for six months

19. Correct the following sentences.

- a. One of my friend met me yesterday.

- b. Anitha is going to park everyday at 6.00 p.m.
- c. Myself went to Hyderabad last month.
- d. If you read well, you get the first rank.
- e. There was many students in the hall.
- f. Prasad wants to quickly write the examination.
- g. We ran fastly to catch the bus.
- h. Pallavi prefers milk than coffee.
- i. When did Kamala went to Hyderabad?
- j. Harika returned back my book.

Group 2

2 X 10 = 20

Instructions: 1. Answer any two of the following questions.

2. Each question carries ten marks.

20. Write a report to your Principal on the industrial visit by you to BHEL, Patancheruvu, Hyderabad.

21. Read the following paragraph and make notes:

In 1920, the Congress meeting was held at Nagpur under the leadership of Gandhiji. It was attended by 15000 delegates and the Congress Constitution was amended and resolutions were taken to fight Swaraj by nonviolent methods and undo the injustice done to Punjab and Turkey.

This movement was called Non-Cooperation Movement. Renunciation of honorary titles like 'Sir' given by British, boycott of legislatures, schools and colleges, courts, tendering resignation to government jobs nonpayment of taxes to government were the important programmes of this movement. Gandhi returned his Kaiser-i-Hind title in August, 1920. There were strikes, hartals and burning of foreign goods all over the country. Many Indian were killed in firings and many other were jailed.

In Kerala, a rebellion broke out by Moplah peasants and it was suppressed brutally. Though Gandhiji warned the people many times not to resort to violent methods, on 5th February, 1922 in Chauri-Chaura in Uttar Pradesh people resorted to violence. When policemen opened fire on peaceful demonstrations, the angry people set ablaze the police

station and 22 policemen were killed. Gandhiji stopped the movement because it lost its nonviolent nature. On 10th March, 1922 Gandhiji was arrested for six years.

22. Read the following passage and answer the questions that follow:

Subhas Chandra Bose was born in a Bengali Kayasth family on January 23, 1897 in Cuttack (Odiya Bazaar), Orissa, to Janakinath Bose, and Prabhavati Devi. He was the ninth child of 14. He studied in an Anglo school at Cuttack (now known as Stewart School) until standard 6. He then shifted to Ravenshaw Collegiate School of Cuttack. From there he went to the prestigious Presidency College where he studied briefly. His nationalistic temperament came to light when he was expelled for assaulting Professor Oaten for his anti-India comments.

His high score in the Civil Service examinations meant an almost automatic appointment. He then took his first conscious step as a revolutionary and resigned the appointment on the premise that the best way to end a government is to withdraw from it. At the time, Indian nationalists were shocked and outraged because of the Amritsar massacre and the repressive Rowlatt legislation of 1919. Returning to India, Bose wrote for the newspaper Swaraj and took charge of publicity for the Bengal Provincial Congress Committee. His mentor was Chittaranjan Das, spokesman for aggressive nationalism in Bengal. Bose worked for Das when the latter was elected mayor of Calcutta in 1924. In a roundup of nationalists in 1925, Bose was arrested and sent to prison in Mandalay, where he contracted tuberculosis.

Questions:

- i. When and where was Subhas Chandra Bose?
- ii. Who was his mentor?
- iii. How did Subhas Chandra Bose participate in National Movement after coming back to India?
- iv. Why didn't he join civil Services?
- v. What is the synonym of 'aggressive'?

Department of Technical Education - TELANGANA
State Board of Technical Education and Training – HYDERABAD

Course Title : ENGINEERING MATHEMATICS	Course Code : 18LT-202F
SEMESTER : II	Course Group : COMMON
Teaching Scheme (L : T : P) : 36 :24 : 0 (in Periods)	Credits : 3 Credits
Type of Course : Lecture + Assignments	Total Contact Periods : 60
CIE : 60 Marks	SEE : 40 Marks
Programme : Common to all Engineering Diploma Programmes	

Pre requisites

This course requires the basic knowledge of Algebra, Trigonometry in Mathematics at Secondary school level and Basic Engineering Mathematics at Diploma 1st Semester level

Course Outcomes: COs

At the end of the course, the student will have the ability to:

CO 1	Formulate the equations of Straight Line , Circle and Conic Sections
CO 2	Evaluate the Limits of different Functions
CO 3	Determine the Derivatives of Various Functions
CO 4	Find the Successive Derivatives and Partial Derivatives of Functions
CO 5	Use Differentiation in Geometrical and Physical Applications
CO 6	Find Maxima and Minima.

Course Contents:

Co-ordinate geometry

Unit – I

Duration: 10 Periods (L: 6.0 – T:4.0)

1. **Straight lines:** Write the different forms of a straight line – point slope form, two point

form, intercept form, normal form and general form - Find distance of a point from a line, acute angle between two lines, intersection of two non-parallel lines and distance between two parallel lines - perpendicular distance from a point to a line - Solve simple problems on the above forms

2. **Circle:** Define locus of a point, circle and its equation. Find equation of the Circle given (i) Centre and radius, (ii) two ends of a diameter (iii) Centre and a point on the circumference (iv) three non collinear points and (v) Centre and tangent equation - general equation of a circle - finding Centre, radius - tangent, normal to circle at a point on it - simple problems.

Unit – II

Duration: 8 Periods (L: 4.8 – T:3.2)

3. **Conic Sections:** Define a conic section, focus, directrix, eccentricity, axes and latus rectum – Find equation of a conic when focus directrix and eccentricity are given. Properties of parabola, ellipse and hyperbola - standard forms with Vertex (Centre) at the Origin and Axis (Axes) along Co – Ordinate Axes only – Simple Problems.

Differential Calculus

Unit-III

Duration: 12 Periods (L: 7.2 – T:4.8)

4. **Functions & Limits :** Concept of Limit- Definition- Properties of Limits and Standard Limits (

without proof) $-\lim_{x \rightarrow a} \frac{x^n - a^n}{x - a}, \lim_{x \rightarrow 0} \frac{\sin x}{x}, \lim_{x \rightarrow 0} \frac{\tan x}{x}, \lim_{x \rightarrow 0} \frac{a^x - 1}{x}, \lim_{x \rightarrow 0} \frac{e^x - 1}{x}, \lim_{x \rightarrow 0} (1+x)^{\frac{1}{x}},$

$\lim_{x \rightarrow \infty} \left(1 + \frac{1}{x}\right)^x$ - Simple Problems . Evaluate the limits of the type $\lim_{x \rightarrow l} \frac{ax^2 + bx + c}{\alpha x^2 + \beta x + \gamma}$ and

$$\lim_{x \rightarrow \infty} \frac{f(x)}{g(x)}$$

5. **Differentiation – I :** Concept of derivative - definition from first principle as

$\lim_{h \rightarrow 0} \frac{f(x+h) - f(x)}{h}$ - different notations - derivatives of elementary functions like x^n

, a^x , e^x , $\log x$, $\sin x$, $\cos x$, $\tan x$, $\sec x$, $\operatorname{cosec} x$ and $\cot x$. Derivatives of sum, product, quotient, scalar multiplication of functions - problems. Derivative of function of a function (Chain rule) with illustrative examples such as

$$(i) \sqrt{t^2 + \frac{2}{t}} \quad (ii) x^2 \sin 2x \quad (iii) \frac{x}{\sqrt{x^2 + 1}} \quad (iv) \log(\sin(\cos x)).$$

Unit – IV

Duration: 12 Periods (L: 7.2 – T:4.8)

6. **Differentiation – II:** Derivatives of inverse trigonometric functions, derivative of a function with respect to another function, derivative of parametric functions, derivative of hyperbolic, implicit functions, logarithmic differentiation – problems in each case. Higher order derivatives - examples – functions of several variables – partial differentiation, Euler’s theorem-simple problems.

Applications of Derivatives:

Unit – V

Duration: 8 Periods (L: 4.8 – T:3.2)

7. **Geometrical Applications: Geometrical** meaning of the derivative, equations of Tangent and normal to a curve at any point on the curve. Lengths of tangent, normal, sub tangent and subnormal to the curve at any point on it. Angle between two intersecting curves - problems.

Unit – VI

Duration: 10 Periods (L: 6 – T:4)

8. **Physical Applications: Physical** applications of the derivative – Explain the derivative as a rate of change in distance-time relations to find the velocity and acceleration of a moving particle with examples. Explain the derivative as a rate measure in the problems where the quantities like volumes, areas vary with respect to time- illustrative examples– Simple Problems.
9. **Maxima & Minima: Applications** of the derivative to find the extreme values – Increasing and decreasing functions, finding the maxima and minima of simple functions - problems leading to applications of maxima and minima.

References

1. Co - Ordinate Geometry – by S.L. Loney
2. Thomas Calculus, Pearson Addison – Wesley Publications
3. Calculus – I by Shanti Narayan and Manicavachagam Pillai, S.V Publications.
4. NCERT Mathematics Text Books Of Class XI, XII.

5. Intermediate Mathematics Text Books (Telugu Academy)

Suggested E-Learning references

1. www.freebookcentre.net/mathematics/introductory-mathematics-books.html

2.E-books:www.mathebook.net

Specific Learning Outcomes

Coordinate Geometry

Unit – I

1.0 Solve the problems on Straight lines

- 1.1 Write the different forms of a straight line – point slope form, two point form, intercept form, normal form and general form
- 1.2 Solve simple problems on the above forms
- 1.3 Find distance of a point from a line, acute angle between two lines, intersection of two non-parallel lines and distance between two parallel lines.

2.0 Solve the problems on Circles

- 2.1 Define locus of a point, circle and its equation.
- 2.2 Find the equation of a circle given
 - (i) Centre and radius
 - (ii) Two ends of a diameter
 - (iii) Centre and a point on the circumference
 - (iv) Three non collinear points
 - (v) Centre and tangent
- 2.3 Write the general equation of a circle and find the centre and radius.
- 2.4 Write the equation of tangent and normal at a point on the circle.
- 2.5 Solve the problems to find the equations of tangent and normal.

Unit – II

3.0 Appreciate the properties of Conics in engineering applications

- 3.1 Define a conic section.
- 3.2 Understand the terms focus, directrix, eccentricity, axes and latus rectum of a conic with illustrations.
- 3.3 Find the equation of a conic when focus, directrix and eccentricity are given
- 3.4 Describe the properties of Parabola, Ellipse and Hyperbola

3.5 Solve problems in simple cases of Parabola, Ellipse and Hyperbola.

Differential Calculus

UNIT - III

4.0 Use the concepts of Limit for solving the problems

4.1 Understand the concept of limit and meaning of $\lim_{x \rightarrow a} f(x) = l$ and state the properties of limits.

4.2 Mention the Standard limits $\lim_{x \rightarrow a} \frac{x^n - a^n}{x - a}$, $\lim_{x \rightarrow 0} \frac{\sin x}{x}$, $\lim_{x \rightarrow 0} \frac{\tan x}{x}$, $\lim_{x \rightarrow 0} \frac{a^x - 1}{x}$, $\lim_{x \rightarrow 0} \frac{e^x - 1}{x}$, $\lim_{x \rightarrow 0} (1+x)^{\frac{1}{x}}$, $\lim_{x \rightarrow \infty} \left(1 + \frac{1}{x}\right)^x$ (All without proof).

4.3 Solve the problems using the above standard limits

4.4 Evaluate the limits of the type $\lim_{x \rightarrow l} \frac{ax^2 + bx + c}{\alpha x^2 + \beta x + \gamma}$ and $\lim_{x \rightarrow \infty} \frac{f(x)}{g(x)}$

5.0 Appreciate Differentiation and its meaning in engineering situations

5.1 State the concept of derivative of a function $y = f(x)$ – definition, first principle as

$\lim_{h \rightarrow 0} \frac{f(x+h) - f(x)}{h}$ and also provide standard notations to denote the derivative of a function.

5.2 State the significance of derivative in scientific and engineering applications.

5.3 Find the derivatives of elementary functions like x^n , a^x , e^x , $\log x$, $\sin x$, $\cos x$, $\tan x$, $\sec x$, $\operatorname{cosec} x$ and $\cot x$ using the first principles.

5.4 Find the derivatives of simple functions from the first principle.

5.5 State the rules of differentiation of sum, difference, scalar multiplication, product and quotient of functions with illustrative and simple examples.

5.6 Understand the method of differentiation of a function of a function (Chain rule) with illustrative examples such as

$$(i) \sqrt{t^2 + \frac{2}{t}} \quad (ii) x^2 \sin 2x \quad (iii) \frac{x}{\sqrt{x^2 + 1}} \quad (iv) \log(\sin(\cos x)).$$

Unit – IV

6.0 Appreciate Differentiation and its meaning in engineering situations

6.1 Find the derivatives of Inverse Trigonometric functions and examples.

- 6.2 Understand the method of differentiation of a function with respect to another function and also differentiation of parametric functions with examples.
- 6.3 Find the derivatives of hyperbolic functions.
- 6.4 Explain the procedures for finding the derivatives of implicit function with examples.
- 6.5 Explain the need of taking logarithms for differentiating some functions with examples like $[f(x)]^{g(x)}$.
- 6.6 Explain the concept of finding the higher order derivatives of second and third order with examples.
- 6.7 Explain the concept of functions of several variables, partial derivatives and difference between the ordinary and partial derivatives with simple examples.
- 6.8 Explain the definition of Homogenous function of degree n
- 6.9 Explain Euler's theorem for homogeneous functions with applications to simple problems.

Applications of Differentiation

UNIT - V

7.0 Understand the Geometrical Applications of Derivatives

- 7.1 State the geometrical meaning of the derivative as the slope of the tangent to the curve $y=f(x)$ at any point on the curve.
- 7.2 Explain the concept of derivative to find the slope of tangent and to find the equation of tangent and normal to the curve $y=f(x)$ at any point on it.
- 7.3 Find the lengths of tangent, normal, sub-tangent and sub normal at any point on the curve $y=f(x)$.
- 7.4 Explain the concept of angle between two curves and procedure for finding the angle between two given curves with illustrative examples.

Unit – VI

8.0 Understand the Physical Applications of Derivatives

- 8.1 Explain the derivative as a rate of change in distance-time relations to find the velocity and acceleration of a moving particle with examples.
- 8.2 Explain the derivative as a rate measurer in the problems where the quantities like volumes, areas vary with respect to time- illustrative examples.

9.0 Use Derivatives to find extreme values of functions

- 9.1 Define the concept of increasing and decreasing functions.
- 9.2 Explain the conditions to find points where the given function is increasing or decreasing with illustrative examples.

- 9.3 Explain the procedure to find the extreme values (maxima or minima) of a function of single variable - simple problems yielding maxima and minima.
- 9.4 Solve problems on maxima and minima in applications like finding areas, volumes, etc.

Suggested Student Activities

1. Student visits Library to refer Standard Books on Mathematics and collect related material.
2. Quiz
3. Group discussion
4. Surprise tests
5. Seminars
6. Home assignments.

Course Content and Blue Print of Marks for SEE – 202F

UNIT No.	UNIT NAME	Periods	Questions for SEE			Marks weightage	Weightage %
			R	U	A		
I	Straight lines	4	1			2	8
	Circles	6	1	1		7	
II	Conic Sections	8		1		10	9
III	Functions & Limits	6	1			2	8
	Differentiation – I	6	1	1		7	
IV	Differentiation – II	12		1		10	9
V	Geometrical Applications	8	3		1+2	31 (6+5+20)	28.5
VI	Physical Applications	5	2		2+1	24 (4+5+5+10)	37.5
	Maxima & Minima	5	1		1+1	17 (2+5+10)	
	Total	60	10	4	8	110	100

R – Remembering: 20 M U – Understanding : 30 M A -- Application : 60 M

DISTRIBUTION OF QUESTIONS/MARKS FOR SEMESTER MID/END EXAMINATION – 202F

Module	Unit No.	No. of Periods	NUMBER OF QUESTIONS TO BE CONSIDERED										UNIT WISE WEIGHTAGE	(MS+EE) WEIGHTAGE
			R		U		A		MARKS WEIGHTAGE					
			MID	END	MID	END	MID	END	MS-I	MS-II	MS-III	END EXAM		
PART-A	I	10	3	2	2	1	2	0	36	-	-	9	45	74
	II	8	2	0	1	0	1	1	19	-	-	10	29	
PART-B	III	12	3	2	1	1	1	0	-	21	-	9	30	74
	IV	12	2	0	2	0	2	1	-	34	-	10	44	
PART-C	V	8	2	1	1	0	1	1	-	-	19	12	31	72
	VI	10	3	0	2	1	2	0	-	-	36	5	41	
TOTAL		60	15	5	9	3	9	3	55	55	55	55	220	220
												110		

LEGEND	R: Remembering
	U: Understanding
	A: Applying

Question Paper Blue Print for SEE

Course: ENGINEERING MATHEMATICS CODE: 18COMMON202F

UNIT No./NAME		No. of Hours	PART – A 2 Marks	PART – B 5 Marks	PART– C 10 Marks	Marks weightage	Weightage (%)
I	a).Straight Lines	04	01	----	----	02	8
	b).Circles	06	01	01	----	07	
II	Conic Sections	08	----	----	01	10	9
III	a).Functions & Limits	06	01	----	----	02	8
	b). Differentiation – I (up to Chain rule)	06	01	01	----	07	
IV	Differentiation – II	12	----	----	01	10	9
V	Geometrical Applications	08	03	01	02	31	28.5
VI	a).Physical Applications	05	02	02	01	24	37.5
	b).Maxima and Minima	05	01	01	01	17	
TOTAL		60	10	06	06	110	100
Questions to be Answered			10	04	04	80	

STATE BOARD OF TECHNICAL EDUCATION & TRAINING, TELANGANA

DIPLOMA EXAMINATIONS, MODEL PAPER, II SEMESTER

ENGINEERING MATHEMATICS

TIME: 3 Hours

Max. Marks: 80

PART – A

Marks: 10 X 2 = 20

*NOTE: 1) Answer ALL questions and each question carries Two marks.
2) Answers should be brief and straight to the point and shall not exceed three simple sentences*

1. Find the slope of the curve $y = x^2 + 2x - 1$ at (1, 2)
2. Write the formula to find the equation of Normal to a given curve at a point (x_1, y_1)
3. Write the condition for orthogonality of two intersecting curves?
4. Find the velocity of a particle when $S = t^2 + 3t - 4$ at $t = 1$ sec .
5. Find the acceleration of the particle when $S = t^2 - 6t + 8$ (t is in sec.) at the instant where the velocity is zero.
6. Define increasing and decreasing functions.
7. Find the distance between parallel lines $2x + 3y + 5 = 0$ and $2x + 3y + 9 = 0$
8. Find the centre of the circle $x^2 + y^2 - 6x + 4y - 12 = 0$.
9. Evaluate $\lim_{x \rightarrow 0} \frac{2x^3 - 3x^2 + 1}{9x^2 + 8x + 7}$
10. Differentiate $x^3 + \sin x$ w.r.t. x

PART – B

Group -1

Marks: 2 X 5 = 10

*NOTE: 1) Answer any Two questions and each question carries Five marks
2) The answers should be comprehensive and the criteria for valuation is the content but not the length of the answer.*

11. Find the equations of the tangent and normal to the curve $y = x^2 + 2x - 1$ at (1,2)
12. The volume of spherical balloon is increasing at a rate of 40 cu.cm/sec. Find the rate of increase of its surface area and radius at the instant when its radius is 10 cms.
13. Find the Maximum and Minimum values of $4x^3 - 3x^2 - 18x + 12$

Group -2

Marks : 2 X 5 = 10

*NOTE: 1) Answer any **Two** questions and each question carries **Five** marks*

2)The answers should be comprehensive and the criteria for valuation is the content but not the length of the answer.

14. Find the equation of the Circle passing through the points (0, 0), (a,0) and (0, b).

15. Find $\frac{dy}{dx}$, If $y = \sin(x^2 + 2x + 1)^3$

16. A Circular metal plate expands by heat so that its radius is increasing at the rate of 0.02 cm per second. At what rate its area is increasing when the radius is 20 cm?

PART – C

Group-1

Marks: 2 X 10 = 20

*NOTE:1) Answer any **Two** questions and each question carries **Ten** marks*

2)The answers should be comprehensive and the criteria for valuation is the content but not the length of the answer.

17. Show that the curves $y^2 = 4ax$ and $xy = c^2$ cut each other orthogonally

$$\text{if } c^4 = 32a^4$$

18. An inverted cone has a depth of 10 cm., base radius is 4 cm , water is poured in to the cone at the rate of 1 cc/sec . Find the rate at which the level of water is increasing when the height of the water level is 6 cm .

19 A rectangular sheet of metal of dimensions 8cm X 5 cm , equal squares are cut off from the corners and the flaps are then folded up to form an open rectangular box . Find the side of the square cut off so that the box may be of greater capacity. What is the maximum capacity of the box so made.

Group-2

Marks: 2 X 10 = 20

*NOTE:1) Answer any **Two** questions and each question carries **Ten** marks*

2)The answers should be comprehensive and the criteria for valuation is the content but not the length of the answer.

20. Find the Centre, Vertices, Eccentricity, Foci, Lengths of axes, Length of Latus Rectum,

and equations of Directrices of the Hyperbola $\frac{x^2}{25} - \frac{y^2}{16} = 1$

21 a) Find the derivative of $\log \sin x$ w.r.t $\cot x$

b) Find $\frac{\partial^2 z}{\partial x^2}$ & $\frac{\partial^2 z}{\partial y^2}$, if $z = x^3 + y^3 - 3axy$

22. a) Find the angle between the curves $y = x^2$, $y = 4 - x^2$

b) Find the lengths of the tangent, normal, sub-tangent and sub-normal for the curve

$y = x^3 - 3x^2 - 8x - 2$ at $(3, 4)$

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STATE BOARD OF TECHNICAL EDUCATION & TRAINING, TELANGANA

BOARD DIPLOMA EXAMINATIONS

MID SEM –I, MODEL PAPER, II SEMESTER

ENGINEERING MATHEMATICS

TIME: 1: 30 Hours

Max. Marks: 40

PART – A

Marks: 5 X 2 = 10

NOTE: 1) Answer ALL questions and each question carries Two marks.

2) Answers should be brief and straight to the point and shall not exceed three simple sentences

1. Find the distance between parallel lines $2x + 3y + 5 = 0$ and $2x + 3y + 9 = 0$
2. Find the slope of the straight line $3x + 4y + 9 = 0$
3. Find the radius of the circle $x^2 + y^2 - 6x + 4y - 12 = 0$.
4. Find the focus and length of latus rectum of the parabola $y^2 = 8x$
5. Find the Length of major and minor axes of the Ellipse $\frac{x^2}{16} + \frac{y^2}{9} = 1$

PART – B

Marks: 2 X 5 = 10

NOTE: 1) Answer any Two questions and each question carries Five marks

2) The answers should be comprehensive and the criteria for valuation is the content but not the length of the answer.

6. Find the point of intersection of the lines $x - 3y + 6 = 0$ and $2x + 3y - 10 = 0$.
7. Find the equation of the Circle passing through the points (0, 0), (1,0) and (0,2)
8. Find the equation of the Parabola with focus at (2 , -3) and whose directrix is $3x - 4y + 16 = 0$

PART – C

Marks: 2 X 10 = 20

NOTE: 1) Answer any Two questions and each question carries Ten marks

2) The answers should be comprehensive and the criteria for valuation is the content but not the length of the answer.

- 9a) Find the angle between the lines $2x - y + 3 = 0$ and $x + y - 2 = 0$
- b) Find the equation of the straight line passing through the point (2 , -5) and perpendicular to the line $7x + 2y - 1 = 0$.
- 10 Find the equations of the Tangent and Normal to the Circle $x^2 + y^2 - 6x - 3y - 2 = 0$ at (2 , -2)

11. Find the Centre, Vertices, Eccentricity, Foci, Lengths of axes, Length of Latus Rectum,

and equations of Directrices of the Hyperbola $\frac{x^2}{16} - \frac{y^2}{9} = 1$

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STATE BOARD OF TECHNICAL EDUCATION & TRAINING, TELANGANA

BOARD DIPLOMA EXAMINATIONS

MID SEM –II, MODEL PAPER, II SEMESTER

ENGINEERING MATHEMATICS

TIME: 1: 30 Hours

Max. Marks: 40

PART – A

Marks: 5 X 2 = 10

*NOTE: 1) Answer ALL questions and each question carries Two marks.**2) Answers should be brief and straight to the point and shall not exceed three simple sentences*

1. Evaluate $\lim_{x \rightarrow 0} \frac{2x^3 - 3x^2 + 1}{9x^2 + 8x + 7}$
2. Evaluate $\lim_{x \rightarrow 2} \frac{x^3 - 8}{x - 2}$
3. Differentiate $x^3 + \tan x$ w.r.t. x
4. Find $\frac{d^2y}{dx^2}$, if $y = x^3 + 4x^2 - 8x + 2$
5. Find $\frac{\partial u}{\partial x}$ if $U = x^3 + y^3 + 3axy$

PART – B

Marks: 2 X 5 = 10

*NOTE: 1) Answer any Two questions and each question carries Five marks**2) The answers should be comprehensive and the criteria for valuation is the content but not the length of the answer.*

6. Evaluate ; $\lim_{n \rightarrow \infty} \left(\frac{1+2+3+\dots+n}{n^2} \right)$
7. If $y = \cos x^{\tan x}$, find $\frac{dy}{dx}$
8. Find $\frac{dy}{dx}$ if $x = at^2, y = 2at$.

PART – C

Marks: 2 X 10 = 20

*NOTE: 1) Answer any Two questions and each question carries Ten marks**2) The answers should be comprehensive and the criteria for valuation is the content but not the length of the answer.*

9. Find $\frac{dy}{dx}$, If $y = \log \frac{1+x^2}{1-x^2}$

10. If $y = \sin(\log x)$, prove that $x^2 y_2 + x y_1 + y = 0$.

11. If $U = \sin^{-1} \left(\frac{x^2+y^2}{x+y} \right)$ then prove that $x \frac{\partial u}{\partial x} + y \frac{\partial u}{\partial y} = \tan u$

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CO / PO - MAPPING

CO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	Mapped POs
CO1	3	2	2							3	1,2,3,10
CO2	3	2	2							3	1,2,3,10
CO3	3	2	2							3	1,2,3,10
CO4	3	2	2							3	1,2,3,10
CO5	3	2	2								1,2,3
CO6	3	2	2								1,2,3

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Department of Technical Education
State Board of Technical Education & Training (TS)

Course Title	Applied Physics	Course Code	18LT-203F
Semester	II	Course Group	Core
Teaching Scheme in Pds/Hrs(L:T:P)	30:15:0 Hrs 40:20:00 Pds	Credits	3
Type of course	Lecture+ Assignments	Total Contact Hrs	60Pds
CIE	60 Marks	SEE	40 Marks

Pre requisites: Basic High school science, basic mathematics

Course Objectives: After studying this course, the student will be able to understand and appreciate the role of Engineering Physics in different areas of engineering and technology.

Course outcomes: On successful completion of the course, the student will have the ability to attain below Course outcomes (CO):

Course Outcomes		Linked POs	Teaching Hours
CO 1	Apply knowledge of waves and sound in engineering problems.	PO1,PO2	10
CO 2	Apply knowledge of Simple Harmonic Motion to solve engineering problems	PO1, PO2	10
CO 3	Use modern instruments in engineering	PO1, PO2,PO3, PO4	10

CO 4	Use various magnetic materials in engineering equipments	PO1, PO2,PO3	10
CO 5	Use various electrical measuring instruments as tools in engineering	PO1, PO2,PO3, PO4	10
CO 6	Apply Electronics principles in engineering problems	PO1, PO2,PO3, PO4	10

APPLIED PHYSICS

Course Contents

1. UNIT – 1 WAVES AND SOUND Duration: 10 periods (L:6.0 – T: 4.0)

Wave motion – definition and characteristics – audible range – infrasonic and ultrasonic – longitudinal and transverse waves – examples – Relation between wavelength, frequency and velocity of a wave – derivation –stationary waves- beats - applications of beats - Doppler effect – list of applications – ultrasound and radar in medicine and engineering as special emphasis - echo –definition - applications - relation between time of echo and distance of obstacle –derivation- Reverberation and time of reverberation - Sabine’s formula - Free and forced vibrations - Resonance - Conditions of good auditorium - noise pollution – definition – effects and methods to minimize noise pollution - problems

2. UNIT – 2 SIMPLE HARMONIC MOTION Duration: 10 periods (L:6.0 – T: 4.0)

Periodic motion - Simple Harmonic Motion (SHM)– definition - examples - Conditions for SHM – Projection of circular motion on any diameter of a circle is SHM - Expressions for Displacement, Velocity and Acceleration of a particle executing SHM – derivations - Time period, frequency, amplitude and phase of particle in SHM - Ideal simple pendulum – time period of simple pendulum – derivation - laws of simple pendulum -Seconds pendulum - problems.

3. UNIT – 3 MODERN PHYSICS Duration: 10 periods (L:6.0 – T: 4.0)

Photo electric effect - Einstein’s photo electric equation – Work function and threshold frequency - laws of photo electric effect - applications of photo electric effect – photo cell - concept of Refraction of light - critical angle and total internal reflection - principle of Optical fiber - Applications of optical fiber – LASER – definition and characteristics –

principle of LASER - spontaneous and stimulated emission-population inversion-examples of LASER – Uses.

4. UNIT-4 MAGNETISM

Duration: 10 periods (L:6.0 – T: 4.0)

Magnetic field - magnetic lines of force -properties - Uniform and Non-uniform magnetic field – Magnetic length, pole strength – magnetic induction field strength- definition - Coulomb’s inverse square law of magnetism - expression for moment of couple on a bar magnet placed in a uniform magnetic field – derivation - expression for magnetic induction field strength at a point on the axial line of a bar magnet -derivation- Dia, Para and Ferro magnetic materials – examples - related problems.

5. UNIT-5 ELECTRICITY AND MEASURING INSTRUMENTS

Duration: 10 periods (L:6.0 – T: 4.0)

Ohm’s law –Ohmic and non ohmic conductors – examples - Temperature dependence of resistance – coefficients of resistance with examples - Specific resistance – units – conductance- moving coil galvanometer - conversion of galvanometer into ammeter and voltmeter with diagram (qualitatively) – Kirchhoff’s current and voltage laws in electricity – Expression for balancing condition of Wheatstone’s bridge – derivation – Meter bridge – working with neat diagram –Superconductivity-definition-superconductors-definition and examples-applications- related problems.

6. UNIT – 6 ELECTRONICS

Duration: 10 periods (L:6.0 – T: 4.0)

Solids – definition – energy bands in solids- valence band, conduction band and forbidden band – Energy band diagram of conductors, insulators and semiconductors – concept of Fermi level - Intrinsic semiconductors - examples - Concept of holes in semiconductors - Doping - Extrinsic semiconductor - P-type and N-type semiconductors - PN Junction diode – Forward Bias and Reverse Bias - Applications of PN diode - Diode as rectifier – principle – principle of Light Emitting Diode and solar cell.

References:

- 1. Engineering Physics by R.K. Gaur, S.L. Gupta, Dhanpatrai Publications, New Delhi.**
- 2. ISC Physics, Book I&II, P. Vivekanandan, DK Banerjee, S Chand, New Delhi.**
- 3. Intermediate Physics, Vol. I&II, Telugu Academy, TS, Hyderabad.**
- 4. Fundamentals of Physics by Halliday and Resnick.**

Specific learning outcomes:

Upon completion of the course the student shall be able to

1. know the concept of Waves and Sound

- 1.1 Define wave. Explain the characteristics of wave (frequency, wavelength, amplitude)
- 1.2 Explain audibility range of sound.
- 1.3 Define infrasonic and ultrasonic sounds.
- 1.4 Define longitudinal and transverse wave motion. Write examples for each. Distinguish between them.
- 1.5 Derive the relation between wavelength, frequency and velocity of wave ($v = n\lambda$)
- 1.6 Define stationary waves.
- 1.7 Explain the phenomenon of beats. List the applications of beats.
- 1.8 Explain Doppler Effect. List the applications of Doppler Effect.
- 1.9 Application of Doppler Effect in medicine and engineering - ultrasound and radar.
- 1.10 Define echo. List the applications of echo.
- 1.11 Derive the relation between time of echo and distance of obstacle.
- 1.12 Explain Reverberation and time of reverberation.
- 1.13 Write Sabine's formula and explain the terms.
- 1.14 Define free and forced vibrations.
- 1.15 Define resonance with examples.
- 1.16 State the conditions of a good auditorium.
- 1.17 Define noise pollution.
- 1.18 List the effects and methods to minimize noise pollution.
- 1.19 Solve related numerical problems.

2. know the concept of Simple Harmonic Motion

- 2.1 Define periodic motion
- 2.2 Define Simple Harmonic Motion (SHM)
- 2.3 List the examples of SHM.
- 2.4 State the conditions of simple harmonic motion
- 2.5 Projection of circular motion on any diameter of a circle is SHM.
- 2.6 Derive the expressions for Displacement, Velocity and Acceleration of a particle executing SHM.
- 2.7 Define the terms time period, frequency, amplitude and phase of particle in SHM
- 2.8 Define Ideal simple pendulum and derive the expression for time period of simple pendulum.
- 2.9 State the laws of simple pendulum.
- 2.10 Define seconds pendulum.
- 2.11 Solve related numerical problems.

3. know the concept of Modern Physics

- 3.1 Explain Photo electric effect.

- 3.2 State Einstein's photo electric equation.
- 3.3 Define terms work function and threshold frequency.
- 3.4 State laws of photo electric effect.
- 3.5 List the applications of photo electric effect.
- 3.6 Define critical angle.
- 3.7 Define Total internal reflection.
- 3.8 State conditions for Total internal reflection
- 3.9 What is Optical fiber? and explain working principle of optical fiber
- 3.10 List the applications of optical fiber.
- 3.11 Explain the principle of LASER.
- 3.12 Define spontaneous and stimulated emission.
- 3.13 Define population inversion.
- 3.14 List the examples of LASER.
- 3.15 List the uses of LASER.

4 know the concept of Magnetism

- 4.1 Define magnetic field.
- 4.2 Define magnetic lines of force.
- 4.3 State the properties of magnetic lines of force.
- 4.4 Define Uniform and Non-uniform magnetic field.
- 4.5 Define the terms magnetic length and pole strength of a bar magnet.
- 4.6 Define magnetic induction field strength.
- 4.7 State and explain Coulomb's inverse square law of magnetism.
- 4.8 Derive the expression for moment of couple on a bar magnet placed in a uniform magnetic field.
- 4.9 Derive the formula for magnetic induction field strength at a point on the axial line of a bar magnet.
- 4.10 Define Dia, Para and Ferro magnetic materials with examples.
- 4.11 Solve related numerical problems.

5 know the concept of Electricity and measuring instruments

- 5.1 State Ohm's law – Define ohmic and non ohmic conductors with examples
- 5.2 Explain temperature dependence of resistance – types of temperature coefficients with examples
- 5.3 Define specific resistance. Write its units.
- 5.4 Define conductance.
- 5.5 Write the formulae for effective resistance in series and parallel combination of resistors.
- 5.6 State and explain Kirchhoff's current and voltage laws in electricity.
- 5.7 Explain moving coil galvanometer.
- 5.8 How a galvanometer is converted to ammeter and voltmeter?
- 5.9 Derive an expression for balancing condition of Wheatstone's bridge with legible sketch.
- 5.10 Explain briefly Meter Bridge with neat diagram.
- 5.11 Define superconductivity.
- 5.12 Define superconductor. Give examples.

- 5.13 List the applications of superconductors.
 5.14 Solve related numerical problems.

6 know the concepts of Electronics and applications

- 6.1 Define solid.
 6.2 Define valence band, conduction band and forbidden band.
 6.3 Explain conductors, insulators and semiconductors on the basis of energy band diagram.
 6.4 Explain the concept of Fermi level.
 6.5 Define intrinsic semiconductors.
 6.6 List the examples for intrinsic semiconductors.
 6.7 Explain the concept of hole in semiconductors.
 6.8 Define doping
 6.9 Define extrinsic semiconductors.
 6.10 Explain P-type and N-type semiconductors.
 6.11 Explain PN Junction Diode (formation only).
 6.12 Explain forward and reverse bias of PN diode (biasing and flow of majority and minority carriers with diagram only)
 6.13 List applications of PN Diode.
 6.14 Explain the principle of diode as a rectifier.
 6.15 Explain working principle of Light Emitting Diode (LED)
 6.16 Explain the working principle of solar cell.

Internal evaluation

Test	Units	Marks	Pattern
Mid Sem 1	1 and 2	20	Part A-5 Short answer questions Part B-2 Essay questions out of 3 Questions Part C-2 Essay questions out of 3 Questions
Mid Sem 2	3 and 4	20	Part A-5 Short answer questions Part B-2 Essay questions out of 3 Questions Part C-2 Essay questions out of 3 Questions
Slip Test 1	1 and 2	5	2 Essay Questions out of 3 Questions
Slip Test 2	3 and 4	5	2 Essay Questions out of 3 Questions
Assignment	1	5	Different group assignments of Higher order Questions that develop problem solving skills and

			critical thinking should be given
Seminars	1	5	
	Total	60	

Suggested Student Activities

1. Student visits Library to refer Text books, reference books and manuals to find their specifications.
2. Student inspects the available equipment in the Physics Lab to familiarize with them.
3. Quiz
4. Seminar
5. Group discussion
6. Surprise test

DISTRIBUTION OF QUESTIONS/MARKS FOR SEMESTER MID/END EXAMINATION

MODULE	UNIT NO	UNIT NAME	Periods	NUMBER OF QUESTIONS TO BE CONSIDERED										UNIT WISE WEIGHTAGE	(MS+EE) WEIGHT AGE
				R		U		A		MARKS WEIGHTAGE					
				MID	END	MID	END	MID	END	MS-I	MS-II	MS-III	END EXAM		
PART-A	1	WAVES AND SOUND	10	3	1	2	1	1	0	26	-	-	7	33	74
	2	SIMPLE HARMONIC MOTION	10	2	1	1	0	2	1	29	-	-	12	41	
PART-B	3	MODERN PHYSICS	10	2	1	1	0	2	1	-	29	-	12	41	74
	4	MAGNETISM	10	3	1	2	1	0	-	26	-	7	33		
PART-C	5	ELECTRICITY AND MEASURING INSTRUMENTS	10	3	0	2	0	1	1	-	-	26	10	36	72
	6	ELECTRONICS	10	2	1	1	1	2	0	-	-	29	7	36	
		TOTAL	60	15	5	9	3	9	3	55	55	55	55	220	220
												110			

Cognitive levels: R=Remember, U=Understand, A=Apply

MODEL QUESTION PAPER (MID SEM-I)
BOARD DIPLOMA EXAMINATION, (C-18)
SECOND SEMESTER, 18 COMMON-203F
APPLIED PHYSICS

Time: $1\frac{1}{2}$ Hours]

[Max Marks: 40

PART-A

Answer **ALL** questions. Each question carries two marks.

5 x 2 = 10

1. Write Sabine's formula and explain terms in it.
2. An observer listens echo from a distant hill in 10 sec. If velocity of sound in air is 340 m/s find distance between observer and hill.
3. Define transverse and longitudinal wave motion.
4. Define SHM and give examples.
5. Find length of seconds pendulum.

PART-B

Answer any **TWO** questions. Each question carries five marks.

2 x 5 = 10

6. Define noise pollution and write methods to reduce noise pollution.
7. Define Doppler effect and write its application.
8. Write conditions for SHM.

PART-C

Answer any **TWO** questions. Each question carries ten marks.

2 x 10 = 20

- 9 (a) Define beats and write its applications. (5)
- (b) Write conditions for good auditorium. (5)
- 10 (a) Derive formula for velocity and acceleration in case of SHM. (6)
- (b) A body under SHM is represented by $y = 10 \sin(6t)$ in meter. Find its maximum velocity and maximum acceleration. (4)
- 11 (a) Define ideal simple pendulum. Find expression for time period in case of simple pendulum. (7)
- (b) State laws of simple pendulum. (3)

MODEL QUESTION PAPER (MID SEM-II)

BOARD DIPLOMA EXAMINATION, (C-18)

SECOND SEMESTER, 18 COMMON-203F

APPLIED PHYSICS

Time: $1\frac{1}{2}$ Hours]

[Max Marks: 40

PART-A

Answer **ALL** questions. Each question carries two marks.

5 x 2 = 10

1. Define photo electric effect.
2. Define threshold frequency.
3. Define magnetic length of a bar magnet.
4. Find the magnetic moment of a bar magnet of length 20 cm and pole strength is 5 A-m.
5. Write any two properties of bar magnet.

PART-B

Answer any **TWO** questions. Each question carries five marks.

2 x 5 = 10

6. Write Einstein's photo electric equation and explain the terms in it.
7. What conditions are required for total internal reflection to take place.
8. State and explain Coulomb's inverse square law.

PART-C

Answer any **TWO** questions. Each question carries ten marks.

2 x 10 = 20

- 9 (a). State the laws photo electric effect. (6)
(b). Write the applications of photo electric effect. (4)
- 10(a). Discuss the expression for moment of a couple on a bar magnet placed in a uniform magnetic field. (6)
(b). A bar magnet of length 20 cm and pole strength 5 A-m makes an angle 30° with a uniform magnetic field of induction 100 tesla. Find the moment of couple on it. (4)
- 11(a). Derive the expression for magnetic induction field strength at a point on the axial line. (7)
(b). Calculate the magnetic induction due to a short bar magnet of magnetic moment

0.5 A m² at a distance of 20 cm on the axial line from the mid point of magnet.
(3)

Semester End Examination marks distribution

	Short Answer	Essay	Marks
Part A	10	0	20
Part B	0	4/6	20
Part C	0	4/6	40
Total	10	8/12	80

BOARD DIPLOMA EXAMINATION, (C-18)

MODEL PAPER

SECOND SEMESTER EXAMINATION

APPLIED PHYSICS

Time: 3 Hours]

[Max Marks: 80

PART-A

10 x 2 = 20

Instructions: (1) Answer **ALL** questions.
 (2) Each question carries **TWO** marks.

1. Define terms reverberation and reverberation time.
2. Define terms time period and frequency in case of SHM.
3. Write Einstein Photo electric equation. Explain terms involved in it.
4. Define uniform magnetic field and non uniform magnetic field.
5. Define ohmic and non ohmic conductors.
6. Define specific resistivity.
7. Define super conductors and give examples.
8. Define conduction band and valence band.
9. Define intrinsic and extrinsic semi conductors.
10. Write applications of PN Diode.

PART-B

GROUP-1

Answer any TWO questions

2 x 5 = 10

11. Define echo. Derive formula for minimum distance to listen echo. (1+4)
12. Define noise pollution. Write four bad effects of noise pollution. (1+4)
13. Derive expression for couple acting on a bar magnet placed inside a uniform magnetic field. (5)

GROUP-2

Answer any TWO questions

2 x 5 = 10

14. Explain conversion of galvanometer into ammeter and voltmeter with the help of diagrams. (2+3)
15. Draw energy band diagrams for conductors, insulators and semi conductors.
16. Explain the working of solar cell.

PART-C

GROUP-1

Answer any TWO questions

2 x 10 = 20

- 17 (a) Derive expression for time period in case of simple pendulum. (7)
(b) Find the length of seconds pendulum on the surface of moon (g on the moon = $1/6$ th of g on the earth) (3)
- 18 (a) Define Dia, para and ferro magnetic materials. (3)
(b) Derive formula for magnetic induction field strength at a point on the axial line of bar magnet. (7)
- 19 (a) State and explain Kirchhoff's law. (6)
(b) Two wires of same material are having lengths in the ratio 2:3 and radii 1:2. Find the ratio of their resistances. (4)

GROUP-2

Answer any TWO questions

2 x 10 = 20

- 20 (a) Derive an expression for balancing condition of Wheatstone's bridge. (7)
(b) Three currents 1 mA, 3 mA and x mA are flowing towards a junction and two currents 2 mA and 3 mA are flowing away from the junction. Find the value of x . (3)
- 21 (a) What is doping? Explain formation of P-type and N-type semi conductors. (7)
(b) Explain principle of diode as rectifier. (3)
- 22 (a) What is PN diode? Draw diagrams for forward and reverse bias. (6)
(b) Explain the working principle of Light Emitting Diode (LED) (4)

Department of Technical Education
State Board of Technical Education & Training (TS)

Course Title: Engineering Chemistry and Environmental Studies	Course Code : 18LT-204F
Semester : Semester II	Course Group : core
Teaching Scheme in hours (L:T:P) : 30:15:00	Credits : 3
Type of course : Lecture + Assignments	Total Contact Hours : 60 periods
CIE : 60 Marks	SEE : 40 Marks

Prerequisite:

Basic knowledge of chemistry in secondary education.

Course Objectives: After studying this course the student will be able to understand and appreciate the role of Chemistry and environmental studies in different spheres of industries.

Course Outcome:

On successful completion of the course, the students will have ability to attain below Course Outcomes (CO):

CO	Course outcome	CL	Linked PO	Teaching periods
CO1	Understand and explain the different metallurgical processes, alloys and applications of alloys	R/U/A	1,2,9,10	10
CO2	Understand and explain corrosion and preventive methods of corrosion	R/U/A	1,2,9,10	10
CO3	List out the different methods of preparation and industrial uses of plastics, rubber and fibers. vulcanization of rubber and its applications	R/U/A	1,2,5,6,7,10	12
CO4	Classify the fuels and explain basic terms of fuel, outline the composition and industrial use of gaseous fuels	R/U/A	1,2,10	8
CO5	Under stand and Explain Galvanic cell, emf of cell - electro chemical series-Applications of Galvanic cells, batteries and cells and distinguish the Galvanic and electrolytic cell	R/U/A	1,2,5	12

CO6	Explain the causes, effects and controlling methods of air and water pollutions.	R/U/A	1,2,5,7	8
	Total Periods			60

Legends: R = Remember, U= Understand, A = Apply

ENGINEERING CHEMISTRY AND ENVIRONMENTAL STUDIES **COURSE CONTENTS**

UNIT - I: Metallurgy: (10 periods)

Characteristics of Metals - distinguish between Metals and Non Metals- Ore, Gangue, Flux, Slag - Concentration of Ore -Froth floatation - Methods of Extraction of crude Metal - Roasting, Calcination, Smelting – Alloys-purpose of making alloys - Composition of Brass, German silver, Nichrome, Stainless steel and Duralumin

UNIT – II: Corrosion: (10 periods)

Introduction - factors influencing the rate of corrosion - electrochemical theory of corrosion - composition, stress and concentration cells- rusting of iron and its mechanism - prevention of corrosion - coating methods, Paints-constituents and characteristics of paints -cathodic protection.

UNIT – III: Polymers: (12 periods)

Introduction - polymerization - types of polymerization - addition, condensation with examples - plastics - types of plastics - advantages of plastics over traditional materials - Disadvantages of using plastics - preparation and uses of the following plastics: 1. Polythene 2. PVC 3. Teflon 4. Polystyrene 5. Urea formaldehyde 6. Bakelite - Rubber - Elastomers –Preparation of Butyl rubber, Buna-s, Neoprene rubber and their uses-Fibres-Preparation and uses of fibres-Nylon 6,6- Polyester.

UNIT – IV: Fuels: (8 periods)

Definition and classification of fuels- characteristics of good fuel - Calorific value-HCV and LCV- Calculation of oxygen required for combustion of methane and ethane - composition and uses of gaseous fuels- a) water gas b) producer gas, c) natural gas, d) coal gas, e) Bio gas and f) acetylene.

UNIT – V: Electro Motive Force (12 periods)

Galvanic cell – standard electrode potential – electro chemical series -emf of cell- Batteries-Types of batteries-Fuel cells.

UNIT – VI: ENVIRONMENTAL STUDIES-II : (8 periods)

Introduction- classification of air pollutants based on origin and states of matter-Air pollution - causes-Effects - control methods - Water pollution - causes - effects - control measures.

Reference Books :

1. Engineering chemistry – Jain & Jain – Dhanpat Rai Publishing Company.
2. A Text book of Engineering Chemistry – S.S.Dara – S.Chand Publications.
3. Environmental Studies – A.K.De.
4. Environmental Studies, R. Rajagopalan, 2nd Edition, 2011, Oxford University Press
5. Intermediate Chemistry I and II – Telugu Academy TS

Specific Learning Outcomes:

Upon completion of the course, the student will have ability to

UNIT – I: METALLURGY

- 1.1. List the Characteristics of Metals.
- 1.2. Distinguish between Metals and Non Metals
- 1.3. Define the terms 1.Mineral, 2.Ore, 3. Gangue, 4. Flux and 5.Slag
- 1.4. Describe Froth Floatation method of concentration of ore.
- 1.5. Describe the methods involved in extraction of crude metal- Roasting, Calcination and Smelting.
- 1.6. Define an Alloy
- 1.7. Explain the purpose of making of alloys
- 1.8. Write the Composition of the following alloys:1.Brass, 2.German silver, 3.Nichrome
4. Stainless steel, 5. Duralumin
- 1.9. List the uses of following Alloys: Brass, German silver, Nichrome, Stainless steel,
Duralumin

UNIT – II: CORROSION

- 2.1. Define the term corrosion
- 2.2. Explain the Factors influencing the rate of corrosion
- 2.3. Explain the concept of electrochemical theory of corrosion
- 2.4. Describe the formation of a) composition cell, b) stress cell c) concentration cell
- 2.5. Define rust and explain the mechanism of rusting of iron with chemical reactions.
- 2.6. Explain the methods of prevention of corrosion:
 - a) Protective Coatings i) Metallic coatings (Anodic and cathodic coatings) ii) Inorganic coatings iii) Organic coatings, paint, constituent of paint and characteristics of good paint.
 - b) Cathodic protection (Sacrificial anode process and Impressed - voltage process).

UNIT – III: POLYMERS

- 3.1. Explain the concept of polymerization
- 3.2. Describe the methods of polymerization a) addition polymerization
b) condensation polymerization with examples.
- 3.3. Define the term plastic
- 3.4. List the Characteristics of plastics.
- 3.5. State the advantages of plastics over traditional materials
- 3.6. State the disadvantages of using plastics.
- 3.7. Types of plastics with examples.
- 3.8. Distinguish between thermoplastics and thermosetting plastics
- 3.9. Explain the methods of preparation and uses of the following plastics:
 1. Polythene, 2. PVC, 3.Teflon, 4. Polystyrene 5. Urea formaldehyde 6. Bakelite

(only flow chart for Bakelite i.e. without chemical equations).

- 3.10. Define the term natural rubber
- 3.11. Explain preparation of natural rubber
- 3.12. State the structural formula of Natural rubber
- 3.13. List the Characteristics of natural rubber
- 3.14. Explain the process of Vulcanization
- 3.15. List the Characteristics of Vulcanized rubber
- 3.16. Define the term Elastomer
- 3.17. Describe the preparation and uses of the following synthetic rubbers a) Butyl rubber, b) Buna-s and c) Neoprene rubber.
- 3.18. Define fibre.
- 3.19. Explain the preparation and uses of fibres –Nylon 6,6 and Polyester

UNIT – IV: FUELS

- 4.1. Define the term fuel
- 4.2. Classify the fuels based on physical state - solid, liquid and gaseous fuels with examples.
- 4.3. Classify the fuels based on occurrence- primary and secondary fuels with examples.
- 4.4. List the characteristics of a good fuel
- 4.5. Advantages of gaseous fuels
- 4.6. Define Calorific value- HCV and LCV.
- 4.7. Calculate the oxygen required for the combustion of Methane and Ethane fuel gases.
- 4.8. State the composition and uses of the following gaseous fuels:
 - a) water gas, b) producer gas, c) natural gas, d) coal gas, e) Bio gas and f) acetylene

UNIT – V: Electro Motive Force

- 5.1. Define Galvanic cell
- 5.2. Explain the construction and working of Galvanic cell
- 5.3. Distinguish between electrolytic cell and galvanic cell
- 5.4. Explain standard electrode potential
- 5.5. Explain standard hydrogen electrode
- 5.6. Define electrochemical series and explain its significance.
- 5.7. Define and explain emf of a cell.
- 5.8. Solve the numerical problems on emf of cell
- 5.9. Explain Batteries (Cells) and types of batteries with examples –working and applications of Dry cell (Leclanche cell), Lead storage battery, Ni-Cd cell
- 5.10. Explain working and advantages of Fuel cell (Hydrogen - Oxygen Fuel Cell)

UNIT – VI: ENVIRONMENTAL STUDIES-II

- 6.1. Define air pollution
- 6.2. Classify the air pollutants- based on origin and states of matter
- 6.3. Explain the causes of air pollution
- 6.4. Explain the effects of air pollution on human beings, plants and animals
- 6.5. Explain the green house effect - ozone layer depletion and acid rain
- 6.6. Explain the methods of control of air pollution
- 6.7. Define water pollution
- 6.8. Explain the causes of water pollution
- 6.9. Explain the effects of water pollution on living and non living things
- 6.10. Explain the methods of control of water pollution.

Internal evaluation:

Test	Units	Marks	Pattern
Mid Sem 1	1 and 2	20	Part A- 5 Short answer questions Part B- 2 Essay questions out of 3 Questions Part C- 2 Essay questions out of 3 Questions
Mid Sem 2	3 and 4	20	Part A 5 Short answer questions Part B 2 Essay questions out of 3 Questions Part C- 2 Essay questions out of 3 Questions
Slip Test 1	1 and 2	5	2 Essay Questions out of 3 Questions
Slip Test 2	3 and 4	5	2 Essay Questions out of 3 Questions
Assignment	1	5	Different group assignments of Higher order Questions that develop problem solving skills and critical thinking should be given , Group discussion
Seminars	1	5	
	Total	60	

Suggested Student Activities for Induction Program:

	Forenoon	Afternoon
Day1	Registration	Class work as per Time table Chemistry Lab practice classes may be conducted
Day2	Rules and Regulations	
Day3	Getting acquainted with Head and faculty	
Day4	Familiarization with Institutional facilities	
Day5	Interaction with Class teacher and Seniors	
Day6	Introducing the mentor	
Day7	Parent –Teacher meeting	

Suggested Student Activities

- 1.Student visits Library to refer to Text books, reference books and manuals to find their specifications
- 2.Student inspects the available equipment in the Chemistry Lab to familiarize with them.
- 3..Quiz
- 4.Group discussion
5. Seminar

6. Surprise test

E learning links:

<https://iupac.org/>

<https://www.youtube.com>

<https://www.khanacademy.org/>

www.nptel.ac.in

DISTRIBUTION OF QUESTIONS/MARKS FOR SEMESTER- MID/END EXAMINATION OF SEMESTER - II

MO DUL E	UNI T NU MBE R	NAME OF THE UNIT	No. OF PERI ODS	NUMBER OF QUESTIONS TO BE CONSIDERED										UNIT WISE WEIGH TAGE	MID+EN D EXAM WEIGHT AGE
				R		U		A		MARKS WEIGHTAGE					
				MID	END	MID	EN D	MI D	EN D	M S-I	MS -II	MS- III	END EXA M		
PAR T-A	I	Metall urgy	10	3	1	1	1	1.5	0.5	26			12	38	74
	II	Corrosi on	10	2	1	2	0	1.5	0.5	29			7	36	
PAR T-B	III	Polyme rs	12	3	1	2	1	1.5	0.5		29		12	43	74
	IV	Fuels	8	2	1	1	0	1.5	0.5		26		7	31	
PAR T-C	V	Electro Motive Force	12	2	1	1	1	2	0			29	7	36	72
	VI	Enviro nmenta l Studies -II	8	3	0	2	0	1	1			26	10	36	
TOTAL			60	15	5	9	3	9	3	55	55	55	55	220	220
												110			

LEG END	R: Remembering
	U: understanding
	A: Applying

Mid term Examination marks distribution

	Short answer	Essay	Marks
Part A	5	0	10
Part B	0	2/3	10
Part C	0	2/3	20
Total	5	4/6	40

Model Question paper:

**Model Paper for Mid-I
BOARD DIPLOMA EXAMINATION, (C-18)
SECOND SEMESTER, 18 COMMON-204F
CHEMISTRY AND ENVIRONMENTAL STUDIES-I**

Time : 1 ½ Hrs

Total Marks :40Marks

PART-A

Answer **all** questions, each carries **two** marks

5 X 2 = 10

1. Define the terms Mineral and Ore.
2. What is Roasting of Ore? Give example.
3. Define Corrosion.
4. Define alloy.
5. What is paint.

PART-B

Answer any **two** questions, each carries **five** marks

2 X 5 = 10

6. Explain the purpose of making alloys.
7. Mention any five factors influencing the rate of corrosion.
8. Explain the mechanism of rusting.

PART-C

Answer any **two** questions, each carries **ten** marks

2 X 10 = 20

9. (a) Explain froth-floatation process of concentration of ore.
(b) Write any five differences between metals and non-metals.
10. (a) Explain the process of calcination and smelting.
(b) Explain the composition and concentration cells formed during corrosion.
11. (a) Differentiate anodic and cathodic coatings.
(b) Explain the sacrificial anode method of protecting metal from corrosion.

Model Paper for Mid-II

BOARD DIPLOMA EXAMINATION, (C-18) SECOND SEMESTER, 18 COMMON-204F CHEMISTRY AND ENVIRONMENTAL STUDIES-I

Time : 1 ½ Hrs

Total Marks :40Marks

PART-A

Answer **all** questions, each carries **two** marks

5 X 2 = 10

1. Define polymerization.
2. Write any two characteristics of plastic.
3. What are fibres.
4. Define fuel. Give two examples.
5. What is the calorific value of a fuel?

PART-B

Answer any **two** questions, each carries **five** marks

2 X 5 = 10

6. What any five differences between thermoplastics and thermosetting plastics.
7. Write any five characteristics of natural rubber.
8. What are primary and secondary fuels? Give examples.

PART-C

Answer any **two** questions, each carries **ten** marks

2 X 10 = 20

9. (a) Write the preparation method and uses of PVC and Teflon.
(b) Write about vulcanization of natural rubber.
10. (a) Write the preparation methods and uses of buna-s and neoprene.
(b) Write any five advantages of gaseous fuel.
11. (a) Write the composition and uses of water gas and producer gas.
(b) Calculate the volume of oxygen required at STP for complete combustion of one mole of ethane.

Semester End Examination marks distribution

	Short answer	Essay	Marks
Part A	10	0	20
Part B	0		
GROUP – I		2/3	10
GROUP - II		2/3	10
Part C	0		
GROUP – I		2/3	20
GROUP - II		2/3	20
Total	10	8/12	80

Model Paper for SEE
BOARD DIPLOMA EXAMINATION, (C-18)
SECOND SEMESTER, 18 COMMON -204F
(SEMESTER END EXAM)
CHEMISTRY AND ENVIRONMENTAL STUDIES-II

Time : 3 Hrs

Total Marks : 80Marks

PART-A

Answer **all** questions, each carries **two** marks

10 X 2 =20

1. Write any four characteristics of metals.
2. Define corrosion.
3. Define addition polymerization.
4. Define calorific value.
5. Define galvanic cell.
6. Define emf.
7. Define standard electrode potential.
8. Define global warming.
9. Define primary pollutant. Give example.
10. Define water pollution.

PART-B

Group - I

Answer any **two** questions, each carries **five** marks

2 X 5 = 10

11. Distinguish electrolytic cells from galvanic cells.
12. Write any five effects of air pollution on human beings.
13. Write a brief note on depletion of ozone layer.

Group - II

Answer any **two** questions, each carries **five** marks

2 X 5 = 10

14. Write about roasting and calcination.
15. Write any five advantages of plastics.
16. Explain the construction of standard hydrogen electrode.

PART-C

GROUP - I

Answer any **two** questions, each carries **ten** marks

2 X 10 = 20

17. (a) Explain the significance of electro chemical series .

(b) Calculate the emf of the following cell if the standard reduction potentials of Zn and Cu are -0.76 V and +0.34 V respectively.



18. (a) Explain about the working and applications of Ni-Cd cell.
(b) Explain about the working of Hydrogen- Oxygen fuel cell and its advantages.
19. (a) Write a brief note on acid rain.
(b) Write about the role of Cottrell electrostatic precipitator and Zoning of industries in controlling air pollution.

GROUP - I

Answer any **two** questions, each carries **ten** marks 2 X 10 = 20

20. (a) Explain the process of concentration of sulphide ore by froth flotation process.
(b) Explain the impressed voltage method of protecting metal from corrosion.
21. (a) Write the preparation method and uses of nylon 6,6 and polyester.
(b) Calculate the volume of oxygen required at STP for complete combustion of one mole of methane.
22. (a) Write any four effects of water pollution.
(b) Explain any three methods of controlling water pollution.

Department of Technical Education
State Board of Technical Education & Training (TS)

Course Title :	Introductory Post tannage	Course Code	18LT-205C
Semester	II	Course Group	:CORE
Teaching Scheme in Hrs(L:T:P)	60periods	Credits	:3
Methodology	Lecture/ self study	Total Contact Hours	: :45Hrs/60Pds
CIE	60marks	SEE	40 Marks

Pre Requisites:

Pre Requisites: The student should have the knowledge of pretanning operations

Course Outcomes (CO):

On successful completion of the course the student will be able to :

Course outcomes		Linked POs	Teaching Hours
CO1	To List the various other mineral tannages for making Leather	P1,P2,P6	8
CO2	Explain the use of different oils and its tanning procedure	P3,P4,P8	12
CO3	To Explain the various post tanning operation of Leather Manufacture	P2	06
CO4	To understand and explain different machine operations done during Leather processing	P3,P4	14
CO5	To identify the different defects present in the wetblue/ EI Leathers and assort them to different Grades	P3,P8	12
CO6	Able to apply Zirconium , aluminium syntans oils, post tanning operations and make use of them in making Leather	P1,P2,P10	8
			60

COURSE CONTENT:

Unit – I : Other Mineral tannages:

Aluminium tanning – properties of Aluminium tanned Leathers – Zirconium tanning - properties of Zirconium tanned Leathers

Unit II : Miscellaneous tanning

Oil tanning – aldehyde tanning – formaldehyde and Glutaraldehyde tanning- Synthetic tanning – properties of oil, aldehyde, and synthetic tanned leathers.

Unit – III : Post Tanning Operations :

Re-tanning – Neutralisation – Dyeing – Fatliquoring and their objectives and method of performing operations – Finishing of leather – Classification of finishes and their constituents.

Unit – IV : Machine Operations :

Different machine operations in leather manufacture – their importance and method of doing each operations.

Unit – V: Assortment :

Importance of assortment

Unit – VI: Application Different mineral salts & chemicals in post tanning operations

Different methods of tanning, post tanning operations – their importance and method of doing each operations.

SUGGESTED BOOKS FOR REFERENCE:

1. Introduction to the principles of leather manufacture - S.S.Dutta. Indian Leather Technology Association, Calcutta.
2. Practical Leather Technology - J.C. Thorstenson, E. Kreiger Publishing company – Malabar, Florida – 1993.
3. Theory and practice of leather manufacture - K.T. Sarkar, Ajoy Sorcar, Madras.
4. Leather Technician hand book - J.H. Sharphouse Leather producer association, North Hampton – 1995. CLRI Publications, Madras.

SPECIFIC LEARNING OUTCOMES:

On completion of the study of the course the student will be able to make use of

Unit I: Other Mineral Tannages

- 1.1 Aluminium tannages
- 1.1.1 Properties of Aluminium tanned Leather
- 1.2 Zirconium Tannages
- 1.2.1 Properties of Zirconium tanned Leathers

Unit II: Miscellaneous Tannages

- 2.1 Oil Tanning:
 - 2.1.1 Oil tannage.
 - 2.1.2 Different oils used in leather industry.
 - 2.1.3 Specific oil used for tanning.
 - 2.1.4 Properties of oil tanned leather.
 - 2.1.5 Uses of oil tanned leather.
 - 2.1.6 Principle of oil tanning for chamois Leather.

- 2.2 Syntans:
 - 2.2.1 Classification of syntans.
 - 2.2.2 Uses of syntans.
 - 2.2.3 Definition of syntans.

- 2.3 Aldehyde tanning
 - 2.3.1 Formaldehyde tanning Principle, Methods and Properties.
 - 2.3.2 Glutaraldehyde tanning Principle, Methods and Properties.
 - 2.3.3 Uses of Formaldehyde and Glutaraldehyde tanned Leathers.

Unit III: Post Tanning Operations.

- 3.1 Retanning.
 - 3.1.1 The purpose of retanning.
 - 3.1.2 The materials used for retanning.
 - 3.1.3 Definition of semi-chrome leather.
 - 3.1.4 Properties of retanned leathers.

3.2 Neutralisation:

- 3.2.1 Purpose of neutralising leathers.
- 3.2.2 Different chemicals used for neutralisation.
- 3.2.3 Duration of neutralisation.
- 3.2.4 P^H changes during neutralisation.
- 3.2.5 Test for neutralisation.
- 3.2.6 The effect of over neutralisation and under neutralisation.

3.3 Dyeing:

- 3.3.1 Purpose of dyeing leather.
- 3.3.2 Definition of dye and colour.
- 3.3.3 Classification of dyes.
- 3.3.4 The selection of dyes for different leathers.
- 3.3.5 Condition required for dyeing.
- 3.3.6 Different methods of dyeing.
- 3.3.7 Fixation of dye to the leather.

3.4 Fat liquoring:

- 3.4.1 Objective of fat liquoring.
- 3.4.2 Classification of fat liquors.
- 3.4.3 Conditions required for fatliquoring
- 3.4.4 Fixation of fatliquor to the leather.

3.5 Finishing

- 3.5.1 Importance of finishing leather
- 3.5.2 Classification of finishes.
- 3.5.3 Preparation of resin finishes.
- 3.5.4 Preparation of glazed finish.
- 3.5.5 Fixing of finishing materials.
- 3.5.6 Operations involved in resin finishing of leather (list of operations).

Unit IV: Machine Operations:

Objectives of the following machine operations and the machinery used.

- 4.1 Sammying.
- 4.2 Splitting.
- 4.3 Shaving.
- 4.4 Setting

- 4.5 Drying and conditioning
- 4.6 Staking and toggling.
- 4.7 Buffing and snuffing.
- 4.8 Glazing.
- 4.9 Spraying.
- 4.10 Plating, printing, ironing and embossing.
- 4.11 Measuring.

Unit V: Assortment:

- 5.1 Importance of assortment.
- 5.2 Different stages at which assortment to made.
- 5.3 Grades in assortments.

Unit VI: Application Different mineral salts & chemicals in post tanning operations:

- 6.1 the advantages of using zirconium salts
- 6.2 the specialty of oil tanned leathers.
- 6.3 the changes take place in post tanning
- 6.4 the present use of Al. zr, syntans salts in leather making

DISTRIBUTION OF QUESTIONS/ MARKS/ FOR SEMESTER MID/END EXAMINATION

MODULE NO	UNIT NO	UNIT NAME	NO OF PERIODS	NUMBER OF QUESTIONS TO BE CONSIDERED										UNIT WISE WEIGHTAGE	(MS+EE) WEIGHTAGE
				R		U		A		MARKS WEIGHTAGE					
				MID	END	MID	END	MID	END	MS-I	MS-II	MS-III	END EXAM		
Part A	I	Other Mineral Tannages	5	2	1	2	1	1	0	24	-	-	7	31	74
	II	Miscellaneous Tannages	15	3	1	1	0	2	1	31	-	-	12	43	
Part B	III	Post-Tanning Operations	15	8	3	1	1	0	1	-	21	-	12	33	74
	IV	Machine Operations	5	2	1	2	1	2	0	-	34	-	7	41	
Part C	V	Assortment	8	1	0	0	0	2	1	-	-	22	10	32	72
	VI	Application Different mineral salts & chemicals in post tanning operations:	12	4	1	3	1	1	0	-	-	33	7	40	
												55	55		
		TOTAL	60	15	5	9	3	9	3	55	55	110	220	220	

R: Remembering U: Understanding A: Applying

INTERNAL EVALUATION:

Test	Units	Marks	Pattern
Mid Sem 1	1 and 2	20	Part A 5 Short answer questions Part B 2 Essay questions out of 3 Questions Part C
Mid Sem 2	3 and 4	20	Part A 5 Short answer questions Part B 2 Essay questions out of 4 Questions Part C
Slip Test 1	1 and 2	5	2 Essay Questions out of 3 Questions
Slip Test 2	3 and 4	5	2 Essay Questions out of 3 Questions
Assignment	1	5	Different group assignments of Higher order Questions that develop problem solving skills and critical thinking should be given
Seminars	1	5	
	Total	60	

MID TERM EXAMINATION MARKS DISTRIBUTION:

	Short answer Questions	Essay Questions	Marks
Part A	5	0	10
Part B	5	2/3	10
Part C	0	2/3	20
Total	5	4/6	40

SEMESTER END EXAMINATION MARKS DISTRIBUTION:

	Short answer Questions	Essay Questions	Marks
Part A	10	0	20
Part B	0	4/6	20
Part C	-	4/6	40
Total	10	8/12	80

SUGGESTED LIST OF STUDENT ACTIVITIES:

Note: The following activities or similar activities for assessing CIE (IA) for 5 marks (Any one)

1. Each student should do any one of the following type activity or any other similar activity related to the course and before conduction, get it approved from concerned Teacher and HOD.
2. Each student should conduct different activity and no repeating should occur

1	Collect various types of Mineral Tanning Salts
2	Compare the various Colour samples of Dyes
3	Collect various Machine photos by downloading from internet and prepare a report

Dimension	RUBRICS FOR ACTIVITY (5 MARKS)					
	Unsatisfactory	Developing	Satisfactory	Good	exemplary	
	1	2	3	4	5	
COLLECTION OF DATA	Does not collect any information relating to the topic	Collects very limited information; some relate to the topic	Collect much information; but very limited relate to the topic	Collects some basic information; most refer to the topic	Collects a great deal of information; all refer to the topic	
Fulfill team's roles & duties	Does not perform any duties assigned to the team role	Performs very little duties but unreliable.	Performs very little duties	Performs nearly all duties	Performs all duties of assigned team roles	
Shares work equally	Always relies on others to do the work	Rarely does the assigned work; often needs reminding	Usually does the assigned work; rarely needs reminding	Normally does the assigned work	Always does the assigned work without having to be reminded.	
Listen to other Team mates	Is always talking; never allows anyone else to speak	Usually does most of the talking; rarely allows others to speak	Talks good; but never show interest in listening others	Listens, but sometimes talk too much	Listens and speaks a fair amount	

SUGGESTED E- LEARNING REFERENCES:

1. <http://www.iltaonleather.org/jilta.html>
2. <https://www.clri.org/>
3. <http://alcajournal.com/>

Course outcomes		Linked POs	Teaching Hours
CO1	To List the various other mineral tannages for making Leather	P1,P2,P6	8
CO2	Explain the use of different oils and its tanning procedure	P3,P4,P8	12
CO3	To Explain the various post tanning operation of Leather Manufacture	P2	06
CO4	To understand and explain different machine operations done during Leather processing	P3,P4	14
CO5	To identify the different defects present in the wetblue/ EI Leathers and assort them to different Grades	P3,P8	12
CO6			

II- SEMESTER Examination C-18 (MPEC) for 3 Year Diploma in Leather Technology

Time: 3 Hours

18LT – 105C INTRODUCTORY POST TANNAGE

Max.Marks: 80

PART-A

Marks: 10x 2 =20

- Note: 1. Answer **ALL** questions and each question carries **TWO** Marks
2. Answer should be brief and straight to be the point and shall not exceed two simple sentences.
1. Write the properties of Aluminium Tanned Leathers?
 2. Write the uses of Chamois Leathers?
 3. Write the objectives of Shaving?
 4. Write a note on Toggling?
 5. Write the objectives of Assortment?
 6. Mention the Different Grades of wet salted goat Skins?
 7. Write the present use of Zirconium Salts?
 8. Write the application of Aluminium Salts?
 9. How will you get the Leather softness by a machine operation?
 10. How will you cover the Grain defect in post tanning operation?

PART- B

Marks: 4 x 5 =20

- Note: 1. Answer any **FOUR** questions and each question carries **FIVE** Marks
2. The Answer should be comprehensive and the criterion for valuation is the content but not the length of answer.
11. Write the Zirconium Tanning method?
 12. Explain the method of Formaldehyde Tanning?
 13. Explain the assortment of Cow Hides at wet salted stage?
 14. Explain the method of Tanning To get white Leathers?
 15. Briefly explain the method of very soft Leather using oils?
 16. Write a procedure for Getting Two layers of cow wetblue?

PART- C

Marks: 4 x10 =40

- Note: 1. Answer any **FOUR** questions and each question carries TEN Marks
2. The Answer should be comprehensive and the criterion for valuation is the content but not the length of answer.
17. Explain the method of oil Tanning?
 18. Describe the method of Fatliquoring?
 19. Explain the method of Resin Finishing of Upper Leathers?
 20. Explain the different stages of assortment in Leather manufacture?
 21. Explain the Different methods of colouring the Leathers?
 22. Write the sequence of post tanning operation for the manufacture of Full chrome leather from wet blue sheep skins?

Department of Technical Education
State Board of Technical Education & Training (TS)

Course Title :	Advanced Engineering Drawing	Course Code	18LT-206P
Semester	II	Course Group	Core
Teaching Scheme in Hrs (L :P)	18.5 : 19.0	Credits	1.5
Methodology	Lecture + practice	Total Contact Hours :	37.5Hrs /45Pds
CIE	60 Marks	SEE	40 Marks

*****This Course is Common to all Programs of Diploma in Level Offered by State Board of Technical Education – Telangana State**

Prerequisites: Knowledge of Basic Engineering Drawing and Clear visualization and sound pictorial intelligence to learn this course.

This Course is aimed at developing basic graphic skills so as to enable them to use these skills in preparation of engineering drawings, their reading and interpretation.

Course Content and Blue Print of Marks for SEE

Unit No	Unit Name	Hours	Questions to be set for SEE				Marks Weightage	Weightage (%)	
			R	U		A			
				SQ	EQ	SQ			EQ
1	Projection Solids	09	--	2	--	--	1	20	18
2	Sectional Views	06		1	--		1	15	14
3	Auxiliary views	06	--	1	--	--	1	15	14
4	Pictorial drawing	15	--	2	--	--	2	30	27
5	Development of surfaces	09	--	2	--	--	2	30	27
	Total	45		40	--	--	70	110	100

R: Remembering, U: Understanding, A: Applying, SQ: Short Questions, EQ: Essay Questions

SEE Question Paper Pattern:

Maximum Marks: 80, Time: 3 Hours

Part A (Short answer questions):

Consists 8 Short Questions, students have to attempt 6 Questions and Each Question Carries 5 Marks. (6 X 5 = 30 M)

Part B (Essay type answer questions):

Consists 7 Essay type Questions, students have to attempt 5 Questions and Each Question Carries 10 Marks. (5 X 10 = 50 M)

Note:

1. To pass in practical Exam student should acquire 50% marks in both CIE and SEE separately and CIE & SEE put together
2. If the students acquire less than 50% in CIE, accordingly the students have to acquire more than 50% in SEE to get overall 50 % to pass.

Course Outcomes (CO)

Upon successful completion of the course, the students will be able to attain the following Course Outcomes (CO):

Course Outcome	
CO1	Apply the principles of Projection of solids also draw the projections of solids.
CO2	Appreciate the need of Sectional views also draw the sectional views and true sections of the engineering components.
CO3	Escalate the need of auxiliary views and draw the auxiliary views of the given engineering components.
CO4	Gain the knowledge of Isometric views of engineering components also draw the isometric views of given components.

CO5	Grasp the knowledge and draw the development of surfaces of different engineering components
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Course Contents

1.0 Projection of solids (09 Hours)

Projection of regular solids

- (a) Axis perpendicular to one of the planes
- (b) Axis parallel to VP and inclined to HP and vice versa.

2.0 Sectional views (06 Hours)

Need for drawing sectional views – what is a sectional view - Location of cutting plane – Purpose of cutting plane line – Selection of cutting plane to give maximum information (vertical and offset planes) - Hatching – Section of regular solids inclined to one plane and parallel to other plane

3.0 Auxiliary views (06 Hours)

Need for drawing auxiliary views -Explanation of the basic principles of drawing an auxiliary views - explanation of reference plane and auxiliary plane -Partial auxiliary view.

4.0 Pictorial Drawings (15 Hours)

Brief description of different types of pictorial drawing viz., Isometric, oblique, and perspective and their use - Isometric drawings: Iso axis, angle between them, meaning of visual distortion in dimensions - Need for an isometric scale, difference between Isometric scale and ordinary scale difference between Isometric view and Isometric projection - Isometric and non-Isometric lines -Isometric drawing of common features like rectangles, circular shapes, non-isometric lines - Use of box and offset methods.

5.0 Development of Surfaces (09 Hours)

Need for preparing development of surface with reference to sheet metal work -Concept of true length of a line with reference to its orthographic projection when the line is (i) parallel to the plane of projection (ii) inclined to one principal and parallel to the other - Development of simple solids like cubes, prisms, cylinders, cones, pyramid (sketches only) -Types of development: Parallel line and radial line development -Procedure of drawing development, drawings of trays, funnels, elbow pipes and rectangular ducts.

Reference Books

- Engineering Drawing by Kapildev – (Asian Publisher)
- Engineering Drawing by BasantAgarwal& C.M Agarwal - (McGraw-hill)
- Engineering Drawing by N.D.Bhatt. (Charotar Publishing House Pvt. Ltd.)
- A Textbook on Engineering Drawing by P. Kannaiah, K. L. Narayana, K. Venkata Reddy

Suggested Specific Learning Outcomes

1.0 Apply Principles of Projection of solids

- 1.1. Draw the projections of solids to axis of solids parallel to one plane.
- 1.2. Draw the projections of solids to axis of solid inclined to other plane.

2.0 Appreciate the need of Sectional Views

- 2.1 Explain the need to draw sectional views.
- 2.2 Select the section plane for a given component to reveal maximum information.
- 2.3 Explain the positions of section plane with reference planes
- 2.4 Differentiate between true shape and apparent shape of section
- 2.5 Draw sectional views and true sections of regular solids discussed in 6.0
- 2.6 Apply principles of hatching.

3.0 Understand the need of auxiliary views

- 3.1 State the need of Auxiliary views for a given engineering drawing.
- 3.2 Draw the auxiliary views of a given engineering component
- 3.3 Differentiate between auxiliary view and apparent view

4.0 Prepare pictorial drawings

- 4.1 State the need of pictorial drawings.
- 4.2 Differentiate between isometric scale and true scale.
- 4.3 Prepare Isometric views for the given orthographic drawings.

5.0 Interpret Development of surfaces of different solids

- 5.1 State the need for preparing development drawing.
- 5.2 Prepare development of simple engineering objects (cubes, prisms, cylinders, cones, pyramid) using parallel line and radial line method.
- 5.3 Prepare development of surface of engineering components like trays, funnel, 90° elbow & rectangular duct.

CIE Question Paper Pattern and Syllabus

Unit No	Unit Name Hours	Questions to be set for SEE				Marks Weightage	Weightage (%)	
		R	U		A			
			SQ	EQ	SQ			EQ
First Mid Examination								
1	Projection Solids	--	3	--	--	2	35	64
2	Sectional Views	--	2	--		1	20	36

	Total		5			3	55	100
Second Mid Examination								
3	Auxiliary views	--	2	--	--	1	20	36
4	Pictorial drawing	--	3	--	--	2	35	64
	Total		5	--	--	3	55	100

R: Remembering, U: Understanding, A: Applying, SQ: Short Questions, EQ: Essay Questions

CIE Question Paper Pattern:

Maximum Marks: 40, Time: 3 Hours

Part A:

Carries **5** Short Questions, students have to attempt **4** Questions and Each Question Carries **5** Marks. ($4 / 5 \times 5 = 20$ Marks)

Part B:

Carries **3** Essay type Questions, students have to attempt **2** Questions and Each Question Carries **10** Marks. ($2 / 3 \times 10 = 20$ Marks)

Note: Students have to get Minimum 50% of the total (i.e. 20 Marks).

Course Outcome		CL	Linked Program Objectives (PO)
CO1	Apply the principles of Projection of solids also draw the projections of solids.	R / U / A	1, 2, 3, 4, 9, 10
CO2	Appreciate the need of Sectional views also draw the sectional views and true sections of the engineering components.	R / U / A	1, 2, 3, 4, 9, 10
CO3	Escalate the need of auxiliary views and draw the auxiliary views of the given engineering components.	R / U / A	1, 2, 3, 4, 9, 10
CO4	Gain the knowledge of Isometric views of engineering components also draw the isometric	R / U / A	1, 2, 3, 4, 9, 10

	views of given components.		
CO5	Grasp the knowledge and draw the development of surfaces of different engineering components	R / U / A	1, 2, 3, 4, 9, 10

Course-PO Attainment Matrix

Course Name	Program Outcomes (PO)									
	1	2	3	4	5	6	7	8	9	10
Advanced Engineering Drawing	3	3	3	2	--	--	--	--	3	3
Level 3- Highly Addressed, Level 2-Moderately Addressed, Level 1-Low Addressed										

Department of Technical Education
State Board of Technical Education & Training (TS)

Course Title :	Advanced Computer Aided Drafting	Course Code	18LT-207P
Semester	II	Course Group	Core
Teaching Scheme in Hrs(L:T:P)	7.5:0:30	Credits	1.5
Methodology	Tutorials + Practice	Total Contact Hours :	37.5Hrs/45Pds
CIE	60 Marks	SEE	40 Marks

Course Content and Blue Print of Marks for SEE

Unit No	Unit name	Periods	Questions for SEE			Marks Weightage	%Weightage
			R	U	A		
1.	Make a drawing with Layers	6				20	25
2	Create and insert the Blocks and add attributes	9		2			
3	Generate Solid models with different surfaces	15		2		20	25
4.	Create complicated 3D drawings using layers and blocks	15			2	40	50
	Total	45		4	2	80	100

R: Remembering, U: Understanding, A: Applying

SEE Question Paper Pattern:

Maximum Marks: 80, Time: 3Hrs

Part – A: Short answer questions: 4 X 10= 40 Marks

Part - B: Essay Type Questions: 2 X 20= 40 Marks

Pre requisites

This course requires the Basic CAD Skills and concepts of engineering drawing

Course Outcomes

CO1	Significance of Layers. Create and assign properties to layers and modify layers
CO2	Importance and advantages of blocks. Use of standard blocks and user defined blocks
CO3	Need of solid modeling and various methods of solid modeling with different surfaces
CO4	Create and modify 3D drawings

Course Contents

UNIT -1

Duration: 06 periods (T:1 + P:4hrs)

Organize the information on layers

Setting a current layer, layers color, line type, line weight, print style locking and unlocking of layers, the layer visibility and layer printing. Setting of current line type. The loading of additional line types, creating and naming of line type, editing of line type.

UNIT -2

Duration: 9 periods (T:1.5 + P:6hrs)

Use the Blocks, Attributes and External references to manage the drawing blocks

The purpose of a block, creating a block, inserting a block, redefining a block, exploding a block

Attribute

Editing attribute definitions, attaching attribute to blocks. Editing attributes attached to blocks, Extracting attributes information.

UNIT -3

Duration: 15 periods (T:2.5 + P:10hrs)

3.0 Viewing entities in three dimensions

Setting a new viewing direction

Dynamically setting a view direction

3.1 Creation of three-dimensional entities using different methods

Drawing of two dimensional entities in three dimensional space.

Converting two dimensional planar entities into three dimensional entities by applying elevation and thickness.

Converting two dimensional planar entities into three dimensional entities by revolving or extruding.

Creation of three-dimensional faces, rectangular meshes, ruled surface meshes, extruded surface meshes, revolved surface meshes, three dimensional entities such as boxes, Cylinders, Cones, Spheres, wedges, torus, Regions, extruded solids, revolved solids, composite solids, intersect solids.

3.2 Editing in three dimensions

Rotating in three dimensions, Array in three dimensions (Rectangular and polar).

Mirroring in three dimensions, aligning in three dimensions.

3.3 Editing of three dimensional solids

Sectioning and Slicing of solids, hiding, shading and rendering.

3.4 Selection of material from library

Enable the material library, editing materials and material library.

UNIT- 4

Application of above three units

Recommended Books

1. Auto cad by George Omura
2. 4MCAD User Guide- IntellCAD Technology Consortium

Suggested learning outcomes

1.0 Organize the information on layers

- 1.1** Explain the need and importance of Layers.
- 1.2** Practice creating new layer, naming the layer and assigning properties like Layer colour, Line type and line weight and setting the current layer
- 1.3** Practice the locking and unlocking of layers
- 1.4** Practice the Freezing and thawing of layers
- 1.5** Practice the layer visibility and layer printing
- 1.6** Practice the loading of additional line types

2.0 Use the Blocks, Attributes and External references to manage the Drawing

- 2.1** Define a block
- 2.2** Explain the purpose of a block
- 2.3** Practice the creating a block

- 2.4 Practice the inserting a block
- 2.5 Practice the redefining a block
- 2.6 Practice the exploding a block
- 2.7 Define an Attribute
- 2.8 Practice the editing attribute definitions
- 2.9 Practice the attaching attribute to blocks
- 2.10 Edit attributes attached to blocks
- 2.11 Extract attributes information
- 2.12 Define external reference
- 2.13 Practice the Attaching, Removing, and Reloading of external references
- 2.14 Practice the Binding, Clipping and changing the path of external references

3.0 Understand the concepts of 3D

View entities in three dimensions

- 3.0.1 To set a new viewing direction
- 3.0.2 To dynamically set a view direction

3.1 Create three-dimensional entities using different methods

- 3.1.1 Draw two dimensional entities in three dimensional space.
- 3.1.2 Convert two dimensional planar entities into three dimensional entities by applying elevation and thickness.
- 3.1.3 Convert two dimensional planar entities into three dimensional entities by revolving or extruding.
- 3.1.4 Create three-dimensional faces.
- 3.1.5 Create rectangular meshes.
- 3.1.6 Create ruled surface meshes.
- 3.1.7 Create extruded surface meshes.
- 3.1.8 Create revolved surface meshes.
- 3.1.9 Create three dimensional entities such as boxes, Cylinders, Cones, Spheres, wedges, torus, Regions.
- 3.1.10 Create extruded solids.
- 3.1.11 Create revolved solids.
- 3.1.12 Create composite solids.
- 3.1.13 Create intersect solids.

3.2 Editing in three dimensions

- 3.2.1 Rotate in three dimensions
- 3.2.2 Array in three dimensions (Rectangular and polar)
- 3.2.3 Mirror in three dimensions
- 3.2.4 Align in three dimensions

3.3 Edit three dimensional solids

- 3.3.1 Practice Sectioning and Slicing solids
- 3.3.2 Practice hiding, shading and rendering

4 Practice applications on above units

Exercise	Key components
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Layers	<ul style="list-style-type: none"> • Importance of layers • Creation of new layers and controlling properties of layers
Working with blocks	<ul style="list-style-type: none"> • Create, insert and explode a block • Attach attribute to blocks • Edit and extract attributes attached to blocks
View entities in three dimensions	<ul style="list-style-type: none"> • Set a new viewing direction • Set dynamically view direction
Create three-dimensional entities	<ul style="list-style-type: none"> • Create three-dimensional faces • Create rectangular meshes, ruled surface meshes, extruded surface meshes, revolved surface meshes • Create three dimensional entities such as boxes, Cylinders.
Edit in three dimensions	<ul style="list-style-type: none"> • Rotate in three dimensions • Array in three dimensions (Rectangular and polar) • Mirror in three dimensions • Align in three dimensions
Edit three dimensional solids	<ul style="list-style-type: none"> • Practice Sectioning and Slicing solids • Practice hiding, shading and rendering
Practice the selection of material from library	<ul style="list-style-type: none"> • Enable material library • Edit materials and material library

Course Outcome		Cognizant Level	Linked PO	Teaching Hours
CO1	Significance of Layers. Create and assign properties to layers and modify layers	U/A	1,2,3,9,10	15
CO2	Importance and advantages of blocks. Use of standard blocks and user defined blocks	U/A	1,2,3,9,10	9
CO3	Need of solid modeling and various methods of solid modeling with different surfaces	U/A	1,2,3,9,10	6
CO4	Create and modify 3D drawings	A	1,2,3,10	15

**Department of Technical Education
State Board of Technical Education & Training (TS)**

Scheme : **C - 18**
Course Title : **POST TANNING LEATHERS PRACTICE**
Semester : **II**
Course Code : **18LT-208P**
Periods per week : **3.0 pds / 2.5 hrs**
Total periods per semester : **45 periods / 54 hr**

Pre requisites: Basic High school science, basic mathematics, and Introduction to Leather Technology

Course outcomes:

At the end of the course the student will have the ability to

CO 1: Summarize the chrome Tanning method.

CO 2: collect and use different materials used for processing Hides /skins in to tanned Leather.

CO 3: List the sequence of post tanning operations and explain the safety precaution methods while working in the lab..

CO 4: Determine the proportion of chemicals used in finishing and List the and classify the grading pattern

CO and PO Mapping

CO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	MAPPING OF POs WITH CO
CO 1	2	2	3		-	-	-	-	-	-	1,2,3,
CO 2	2				-	3	-	3	-	-	2,6,8
CO 3	2	2	-	3	3	-	-	-	-	-	1,2,4,5
CO 4	3	3	3	-	-	3	-	-	-	-	1,2,3,6

Level 3- Highly Addressed, Level 2-Moderately Addressed, Level 1-Low Addressed.

Course content and Blue Print of Marks for SEE

Unit No	Unit Name	Hours	Question to be set for SEE			Marks Weightage	Weightage (%)
			Identifying	Handling	precision		
1	Chrome Tanning	5	1(10)	1(20)	1(10)		
2	Vegetable Tanning	5	1(10)	1(20)	1(10)		
3	Post Tanning Operations	10	1(20)	1(20)			
4	Finishing of Leather	10	1(10)	1(10)	1(10)		
5	Safety Precautions	5	1(20)	1(10)	1(10)		
6	Assortment	10	1(10)	1(20)	1(10)		
	TOTAL	45					

The student will be given any one experiment from the above units for 40 marks

Course contents

1.0 **Chrome Tanning:**

- 1.1 Prepare the chrome liquor.
- 1.2 Tan the pickled stock.
- 1.3 Basify the chrome.
- 1.4 Check completion of chrome tanning by Boil Test.
- 1.5 Determine the Shrinkage Temperature for hydrothermal stability.

2.0 **Vegetable Tanning:**

- 2.1 Prepare Vegetable Tan Liquor by Leaching Method.
- 2.2 Prepare the pelt for vegetable tanning.
- 2.3 Vegetable tan the skins.
- 2.4 Make Sole Leather by Vegetable tanning.
- 2.5 Beam the vegetable tan leather.

3.0 **Post Tanning Operations:**

- 3.1 Sam and set the leathers for splitting and shaving.
- 3.2 Split the sides to reduce thickness and get different layer.
- 3.3 Shave the leather to reduce the leather to uniform thickness.
- 3.4 Neutralise the chrome-tanned leather.
- 3.5 Retan the leather using vegetable tan and chrome tanning material.
- 3.6 Dye the leather.
- 3.7 Fat liquor the leather and fix the fat liquor.
- 3.8 Dry the leather by hook & by nailing.
- 3.9 Condition the leather by saw dusting.
- 3.10 Stake the leather.
- 3.11 Toggle the leather.
- 3.12 Trim the leather without much wastage.
- 3.13 Buff the leathers.

4.0 **Finishing of Leather:**

- 4.1 Prepare the season for resin finish.
- 4.2 Applying the season by pad and brush.
- 4.3 Spraying the season.
- 4.4 Glazing the protein finish leather.
- 4.5 Plate the resin finish leather.
- 4.6 Roll the vegetable tanned sole leather.
- 4.7 Measure the leather area.

5.0 **Safety Precaution in Leather Processing Units:**

- 5.1 Usage of gloves/boots.
- 5.2 Usage of aprons.
- 5.3 Usage of goggles.
- 5.4 Knowledge of electrical connections.

6.0 **Assortment:**

- 6.1 Assortment grades in wet blue.
- 6.2 Grading in EI stage.
- 6.3 Grading of finished leathers.

References:

1. Practical Leather Technology – J H .Sharphouse
2. C L R I publications

Suggested learning outcomes

Upon the completion of study of this subject the student will be able to Understand and do the various techniques in each of the Operations and perform the tests for their proper completion.

POST TANNING OF LEATHERS PRACTICE		
Name of the Experiment	Competencies	Key Competencies
Chrome tanning.	<ul style="list-style-type: none">• To do chrome for the given pelt	<ul style="list-style-type: none">❖ Study the preparation of chrome liquor❖ Chemicals used for chrome tanning❖ Duration of chrome tanning for diff types of leather❖ Test for completion of chrome❖ Ph changes observed during chrome tanning
Vegetable tanning	<ul style="list-style-type: none">• To do vegetable tanning for the given pelt	<ul style="list-style-type: none">❖ Study the preparation of veg tan liquor❖ Leaching method of tan liquor preparation❖ Duration of vegetable tanning for diff types of leather❖ Test for completion of vegetable tanning❖ Ph changes observed during vegetable

Samming	<ul style="list-style-type: none"> • To do samming for the chrome tanned leather • To do samming for the vegetable tanned leather 	<ul style="list-style-type: none"> ❖ The condition of the sammed leather ❖ The moisture content of the sammed leather ❖ various conditioning methods
Setting	<ul style="list-style-type: none"> • To do setting for the chrome tanned leather • To do setting for the vegetable tanned leather 	<ul style="list-style-type: none"> ❖ The condition of the setted leather ❖ The grain condition of the setted leather ❖ various setting proceeedures
Splitting	<ul style="list-style-type: none"> • To do splitting of chrome tanned leather • To do splitting of the vegetable tanned leather 	<ul style="list-style-type: none"> ❖ The thickness of grain layer ❖ The condition of the split layer ❖ The splitting mechanism
Shaving	<ul style="list-style-type: none"> • To do shaving of chrome tanned leather • To do shaving of the vegetable tanned leather 	<ul style="list-style-type: none"> ❖ The thickness of grain layer ❖ The condition of the shaved leather
Rechroming	<ul style="list-style-type: none"> • To do rechroming of chrome tanned leather • To do semi chroming of the vegetable tanned leather 	<ul style="list-style-type: none"> ❖ Study necessary of chrome liquor ❖ Chemicals used for rechroming ❖ Duration of rechroming for diff types of leather ❖ Test for completion of rechroming ❖ Ph changes observed during rechroming

neutralisation	<ul style="list-style-type: none"> • To do neutralisation f/c leather • To do neutralisation of semi chrome leather 	<ul style="list-style-type: none"> ❖ Purpose of neutralising leathers. ❖ Different chemicals used for neutralisation. ❖ Duration of neutralisation. ❖ P^H changes during neutralisation. ❖ Test for neutralisation. ❖ The effect of over neutralisation and under neutralisation
Retanning	<ul style="list-style-type: none"> • To do retanning f/c leather • To do neutralisation of semi chrome leather 	<ul style="list-style-type: none"> ❖ Purpose of retanning leathers. ❖ Different chemicals used for Retanning. ❖ Duration of Retanning. ❖ P^H changes during Retanning.
Dyeing	<ul style="list-style-type: none"> • To do dyeing of f/c leather • To do dyeing of semi chrome leather 	<ul style="list-style-type: none"> ❖ Purpose of dyeing leathers. ❖ Different dyes used for dyeing. ❖ Duration of dyeing. ❖ P^H changes during dyeing ❖ Fixation of dyes. ❖ Colour matching
Fatliquoring	<ul style="list-style-type: none"> • To do Fatliquoring of f/c leather • To do fatliquoring of semi chrome leather 	<ul style="list-style-type: none"> ❖ Purpose of Fatliquoring leathers. ❖ Different Fatliquoring used for Fatliquoring ❖ Duration of Fatliquoring. ❖ P^H changes during Fatliquoring ❖ Fixation of Fatliquoring. ❖ Names of some fatliquors
Samming	<ul style="list-style-type: none"> • To do samming for the chrome tanned leather • To do samming for the vegetable tanned leather 	<ul style="list-style-type: none"> ❖ The condition of the sammed leather ❖ The moisture content of the sammed leather ❖ various conditioning methods
Setting	<ul style="list-style-type: none"> • To do setting for the chrome tanned leather • To do setting for the vegetable tanned leather 	<ul style="list-style-type: none"> ❖ The condition of the setted leather ❖ The grain condition of the setted leather ❖ various setting procedures

Drying	<ul style="list-style-type: none"> • To do drying for the chrome tanned leather • To do drying for the vegetable tanned leather 	<ul style="list-style-type: none"> ❖ The condition of the dried leather ❖ The grain condition of the leather ❖ Different methods of drying like air drying, drying by nailing, toggle drying
Conditioning	<ul style="list-style-type: none"> • To condition the chrome tanned leather • To condition the vegetable tanned leather 	<ul style="list-style-type: none"> ❖ The condition of the conditioned leather ❖ The moisture content of the conditioned leather ❖ Different methods of conditioning like saw dusting spraying etc.,
Staking	<ul style="list-style-type: none"> • To do staking of chrome tanned leather • To do staking of the vegetable tanned leather 	<ul style="list-style-type: none"> ❖ The change in softness of leather ❖ The condition of the staked leather ❖ Different methods of staking machines
Toggling	<ul style="list-style-type: none"> • To do toggling of chrome tanned leather • To do toggling of the vegetable tanned leather 	<ul style="list-style-type: none"> ❖ The change in softness of toggled leather ❖ The condition of the toggled leather ❖ different methods of toggling procedures
Buffing / snuffing	<ul style="list-style-type: none"> • To do buffing of chrome tanned leather • To do buffing of the vegetable tanned leather 	<ul style="list-style-type: none"> ❖ The change in condition of buffed leather ❖ The various emery papers used for buffing ❖ Observation of the nap of buffed leather ❖ Dust off procedure
Finishing	<ul style="list-style-type: none"> • To prepare the season for resin finish • To prepare the season for protein finish 	<ul style="list-style-type: none"> ❖ The change in condition of buffed leather ❖ The various emery papers used for buffing ❖ Observation of the nap of buffed leather ❖ Dust off procedure
Glazing	<ul style="list-style-type: none"> • To do glazing for protein finished leather 	<ul style="list-style-type: none"> ❖ The change in condition of glazed leather ❖ Observe the gloss& appearance of the glazed leather

Plating	<ul style="list-style-type: none"> To do plating for resin finished leather 	<ul style="list-style-type: none"> The change in condition of plated leather Observe the grain & appearance of the plated leather
Measuring / safety precaution	<ul style="list-style-type: none"> To measure the area of finished leather 	<ul style="list-style-type: none"> Observe the area in square feet Precautions to be observed during the practicals in the tannery
Assortment	<ul style="list-style-type: none"> To know the different stages of assortment Know the Basis of assortment 	<ul style="list-style-type: none"> Assortment grades in wet blue. Grading in E1 stage. Grading of finished leathers Defects observed in finished leathers

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GOVERNMENT INSTITUTE OF LEATHER TECHNOLOGY, H.S.DARGA, HYDERABAD 18LT-208P

II- SEMESTER Examination C-18 (MPEC) for 3 Year Diploma in Leather Technology

Time: 3 Hours

18LT – 208P POST TANNING OF LEATHERS PRACTICE

Max.Marks: 40

Instructions to the Candidate:

(i) Answer any One of the following Questions.

(ii) Record the results on a graph sheet if required , and conclude your observation of the experiment

(iii) Draw the t diagram for illustration ,choose appropriate values when not mentioned in the question

1. Prepare chrome liquor from the given Dichromate salt and record the changes takes place during the reaction?
2. Do one bath method of CromeTanning for the given pickled pelts in the drum and record the physio chemical changes observed during the processes?
3. Collect the vegetable tanning material and prepare vegetable tanning Liquor in Leaching Pits?

4. Do vegetable Tanning in drum for the given pelt?
5. Do the mechanical operations like toggling, hand staking for the given dyed crust and record the observations?
6. Do Rechroming in the drum for the given wetblue and record the changes that takes place during the processes?
7. Do Neutralisation in the drum for the given wetblue and record the changes that takes place during the processes?.
8. Do Dyeing in the drum for the given wetblue and record the changes that takes place during the processes?
9. Do Fatliquoring in the drum for the given wetblue and record the changes that takes place during the processes?
10. Do Resin finishing for the given dyed crust as per given shade and record the various ingredients used in it?
11. Do protein finishing for the given dyed crust as per given shade and record the various ingredients used in it?
12. Do Assortment for the given pack of wetblue skins, E I in to different grades and prepare a report with the reasons for selecting it duly mentioning the defects ?

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**Department of Technical Education
State Board of Technical Education & Training (TS)**

COURSE DESIGN DELIVERY AND ASSESSMENT

Course Title	Applied Science Lab Practice	Course Code	18LT-209P(A)
Semester	II	Course Group	Core
Teaching Scheme in Pds/Hrs(L:T:P)	0:1:2 periods	Credits	1.5/2
Type of course	Tutorial & practical	Total Contact Hrs	22.5Pds
CIE	30 Marks	SEE	20 Marks

Pre requisites: Knowledge of basic concepts of basic High school science, basic mathematics

Course objectives: To provide practical knowledge about the basics of Physics instrumentation and calculations/measurements.

Tutorial: 0.83 Hrs/Experiment:

1. Introduction Physics practical and its importance, safety precautions in maintenance of equipment in the laboratory.
2. Maintenance of apparatus and equipment.
3. Follow of Do's and Don'ts.
4. Maintenance of data in manual and record book.
5. Write the procedure of the experiment before the commencement of each experiment.
6. Strictly following of instructions given from time to time by the lecturer-in-charge.
7. Demonstration of each experiment by the lecturer in charge.

Conduct of an experiment: 3periods/experiment.

Course outcomes:

On successful completion of the course, the student will have ability to:

- 1: Determine the Focal length and focal power of convex lenses using U-V and graphical method.

- 2: Determine the value of acceleration due to gravity using Simple Pendulum and verify with L-T²graph.
- 3: Determine the velocity of sound in air at room temperature .
- 4: Determine the refractive index of a solid using travelling microscope.
- 5: Practice the mapping of magnetic lines of force-locating neutral points.

References:

3. Basic Applied Physics – R.K. Gaur
4. Laboratory manual for class XI and XII - NCERT

PHYSICS PRACTICALS

List of experiments

Semester II

1. Convex lens-Determination of Focal length and focal power using U-V and graphical method.
- 2: Simple Pendulum-Determination of the value of acceleration due to gravity and verify with L-T²graph.
- 3: Resonance apparatus-Determination of velocity of sound in air at room temperature .
- 4: Travelling microscope-Determination of refractive index of a solid.
- 5: Practice the mapping of magnetic lines of force-location of neutral points

Course Delivery:

The course will be delivered through lectures, class room interaction, group discussions, graded exercises, demonstration and practice.

Conduction of experiments: 2 periods/Experiment.

Student must perform experiment individually under the supervision of the lecturer-in charge.

On successful completion of the course, the student will have the ability to attain below Course outcomes (CO):

Course Outcomes	CL	Linked experiments	Linked POs	Teaching Hours

CO 1	Focal length and Focal power of convex lens (Separate & Combination)	U/A		1,2,3,8,9	L:P::1:2
CO 2	Acceleration due to gravity using simple pendulum	U/A		1,2,3,8,9	L:P::1:2
CO 3	Velocity of sound in air – (Resonance method)	U/A		1,2,3,8,9	L:P::1:2
CO 4	Refractive index of solid using traveling microscope	U/A		1,2,3,8,9	L:P::1:2
CO 5	Mapping of magnet lines of force-locating neutral points	U/A		1,2,3,8,9	L:P::1:2
CO 6	Related the answers to the oral questions		Covered in all COs		

Cognitive levels: R=Remember, U=Understand, A=Apply

Scheme of Valuation of SEE		
S.No	Particulars	Marks
1.	Identification of apparatus/equipment/etc	01
2.	Writing procedure	04
3.	Conducting of experiment	10
4.	Results	01
5.	Viva-voce	04
	Total	20

Suggested learning outcomes

Name of the Experiment (No of Periods)	Competencies	Key competencies
1. Focal length and Focal power of convex lens (Separate & Combination)	<ul style="list-style-type: none"> • Fix the object distance • Find the Image distance • Calculate the focal length and power of convex lens 	<ul style="list-style-type: none"> • Calculate the focal length and power of convex lens • Draw u-v and $1/u - 1/v$
2. Simple pendulum – acceleration due to gravity – length of seconds pendulum	<ul style="list-style-type: none"> • Fix the simple pendulum to the stand • Adjust the length of pendulum • Find the time for number of oscillations • Find the time period • Calculate the acceleration due to gravity • Draw I-T and $I-T^2$ graph 	<ul style="list-style-type: none"> • Find the time for number of oscillations • Find the time period • Calculate the acceleration due to gravity • Draw I-T and $I-T^2$ graph
3. Velocity of sound in air –Resonance method	<ul style="list-style-type: none"> • Arrange the resonance apparatus • Adjust the reservoir level for booming sound • Find the first and second resonating lengths • Calculate velocity of sound 	<ul style="list-style-type: none"> • Adjust the reservoir level • Find the first and second resonating lengths • Calculate velocity of sound • Calculate velocity of sound at 0°C
4. Refractive index of solid using traveling microscope	<ul style="list-style-type: none"> • Find the least count of vernier on microscope • Place the graph paper below microscope • Read the scale • Calculate the refractive index of glass slab 	<ul style="list-style-type: none"> • Read the scale • Calculate the refractive index of glass slab

<p>5. Mapping of magnet lines of Force – neutral points</p>	<ul style="list-style-type: none"> • Draw magnetic meridian • Place the bar magnet in NN(North pole of bar magnet pointing North) and NS (South pole of bar magnet pointing North) directions • Draw magnetic lines of force • Locate the neutral points along equatorial and axial lines 	<ul style="list-style-type: none"> • Draw magnetic lines of force • Locate the neutral points along equatorial and axial lines
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**Department of Technical Education
State Board of Technical Education & Training (TS)**

Course Title: Applied Science Lab Practice	Course Code: 18LT-209(B)
Semester: II	Core/Elective:
Teaching Scheme(L:P) : 1:2 periods	Credits: 1.5/2
Type of Course : Lecture & practical	Total Contact Hours: 22.5 periods
CIE : 30 Marks	SEE: 20 Marks

Prerequisite: Knowledge of basic concepts of chemistry of secondary education.

Course Objectives: To provide practical knowledge about the basics of volumetric analysis of chemical compounds.

Course Outcomes: On successful completion of the course, the student will have ability to attain CO:

Course Outcome		CL	Linked PO	Teaching Periods
CO1	Estimate the amount of Mohr's salt in the given solution	U/A	1,2,3,8	L:P ::1:2
CO2	Determination of acidity , alkalinity and pH of given water samples / solutions.	U/A	1,2,3,8	L:P ::3:6
CO3	Estimate the amount of the chlorides in the given solution.	U/A	1,2,3,8	L:P ::1:2
CO4	Relate the answers to the oral questions	U/A		

U = Understand, A = Application

Course Delivery: The course will be delivered through lectures, classroom interaction, group discussion, demonstration and practicals.

Conduction of experiments: Lecture 1 period + Experiment 2 periods.

Student must conduct experiment individually under the supervision of the staff-in-charge.

Tutorial:

1. Introduction of chemistry practical and its importance, safety precautions in maintenance of cleanliness and orderliness of chemicals in the laboratory.
2. Maintenance of apparatus and equipment.
3. Follow of DO's and Don'ts.
4. Maintenance of data in record book.
5. Write the procedure of the experiment before the commencement of each experiment.
6. Strict following of instructions given from time to time by the staff-in-charge.
7. Demonstration of each experiment by the staff in charge.

Course content

Volumetric Analysis: (22.5 Hrs)

Volumetric analysis by Titrimetric Method:

Volumetric Analysis -Titration – Standard Solutions- Concentration of solutions- Indicators- acid base indicators- selection of indicators-endpoint of titration-Neutralization.

List of experiments:

1. Estimation of Mohr's salt by using 0.02M potassium permanganate solution.
2. Determination of acidity of water sample by using 0.02N NaOH solution.
3. Determination of alkalinity of water sample by using 0.02N H₂SO₄ solution.
4. Estimate the chloride content present in water sample by using 0.0141N AgNO₃ solution.
5. Find out the pH of the given solution by using pH meter.

Suggested Learning Outcomes:

Upon completion of the course, the student will have ability to

1. Estimate Mohr's salt by using standard potassium permanganate solution.
2. Determine the partial and total acidity of water sample by using 0.02N NaOH solution.
3. Determine the partial and total alkalinity of water sample by using 0.02N H₂SO₄ solution.
4. Estimate the chloride content present in water sample by using 0.0141N AgNO₃ solution.

- Find out the pH of the given solution by using pH meter.

Reference Books:

- Vogel's Inorganic Qualitative and Quantitative Analysis.
- Practical chemistry by O.P.Pande & others.
- Qualitative and quantitative analysis by Alex.

Scheme of Valuation for MID I & II and SEE		
Sl. No.	Particulars	Marks
1	Identification of apparatus/equipment/chemical compounds/tools/etc.	2
2	Writing Procedure	5
3	Conducting of experiment	4
4	Observation and Results	6
5	Viva-voice	3
Total		20

INFORMATION TECHNOLOGY LAB PRACTICE

Course Title : INFORMATION TECHNOLOGY LAB	Course Code : 18LT-210P
PRACTICE	Course Group : Core
Semester : II	Credits : 3
Teaching Scheme in Hrs (L:T:P) : 0:1:2	Total Contact Hours : 37.5Hrs/45Pds
Type of course : Tutorial + Practical	SEE : 40 Marks
CIE : 60 Marks	

Prerequisites

Knowledge of Computer basics and DOS

Course Outcome

On successful completion of the course, the students will be able to attain below Course Outcome (CO):

Course Outcome		CL	Linked PO	Practical hrs
CO1	Demonstrate skills using spreadsheet software	A	1,2,3,4,8,9,10	15
CO2	Demonstrate skills using presentation software	A	1,2,3,4,8,9,10	15
CO3	Demonstrate skills using database software	A	1,2,3,4,8,9,10	15
		Total Sessions		45

Legends: R = Remember U= Understand; A= Apply and above levels (Bloom's revised taxonomy)

Course Content

Spread Sheet

1. Open MS-Excel and identify the components on the screen
2. Create a Worksheet in MS-Excel and save it in .xls or .xlsx format
3. Inserting column and row in Excel
4. Creation of new worksheet in the existing Excel Book file
5. Generate a Chart using the data in Excel-worksheet
6. Automate calculations in a worksheet using formula
7. Sort and filter data in a worksheet
8. Protecting a worksheet, working with multiple sheets
- 9.

Presentation Software

10. Create a simple Power point presentation for a small topic and saving in .ppt or pptx format
11. Inserting a new slide in the existing PowerPoint file
12. Inserting chart or image in a PowerPoint slide
13. Exercise with animation and sound features in PowerPoint
14. Exercise with Rehearse Timings feature in PowerPoint
15. Exercise in printing the PowerPoint file in (a) Slides (b) Handouts

Database Management System

16. Create a table for given data and save in .mdb or .accdb format
17. Add, Delete and rename fields
18. Use the Primary key field
19. Enter and edit data
20. Use Relationships option

21. Create forms
22. Modify and save forms
23. Create and use queries
24. Sort data
25. Display data
26. Create and print reports

Resources:

1. Computer Fundamentals Concepts, Systems, Application, D.P.Nagapal, S.Chand Publication, RP-2014, ISBN: 81-219-2388-3
2. <http://www.tutorialsforopenoffice.org/>
3. <http://www.libreoffice.org/get-help/documentation/>

Composition of Educational Components:

Questions for CIE and SEE will be designed to evaluate the various educational components (Bloom’s taxonomy) such as:

Sl. No.	Bloom’s Category	%
1	Remembrance	20
2	Understanding	20
3	Application	60

**Mapping Course Outcomes with Program Outcomes:
(Course Outcome linkage to Cognitive Level)**

Course Outcome		Experiment Linked	Linked PO	CL	Practical Sessions
CO1	Demonstrate skills using spreadsheet software	1,2,3,4,5,6,7,8	1,2,3,4,8,9,10	A	15
CO2	Demonstrate skills using presentation software	9,10,11,12,13,14	1,2,3,4,8,9,10	A	15
CO3	Demonstrate skills using database software	15,16,17,18,19,20,21, 22,23,24,25	1,2,3,4,8,9,10	A	15

U-Understanding; A-application/ Analysis; App-Application

Course-PO Attainment Matrix

Level 3- Highly Addressed, Level 2-Moderately Addressed, Level 1-Low Addressed.

Course Delivery

The course will be delivered through tutorial of one hour and one & half hours of hands on practice per week.

Suggested Student Activities:

1. Create a spreadsheet for the class
2. Create power point presentation for a course
3. Create a database for the class

Format for Student Activity Assessment

Internal Assesment

Activity	Marks
Writing the experiment, record evaluation	30
Execution of the given experiment	20
Viva-voce	10
Total	60